Open-Source Molecular Modeling Software in Chemical Engineering Focusing on the Molecular Simulation Design Framework

Peter Cummings¹, Clare McCabe¹, Christopher Iacovella¹, Akos Ledeczi³, Eric Jankowski², Arthi Jayaraman⁵, Jeremy Palmer⁴, Edward Maginn⁵, Sharon Glotzer⁶, Joshua Anderson⁶, Ilja Siepmann⁸, Jeffrey Potoff⁸, Raymond Matsumoto¹, Justin Gilmer¹, Ryan DeFever⁵, Ramanish Singh⁷, and Brad Crawford⁸

¹Vanderbilt University
²Boise State University
³University of Delaware
⁴University of Houston
⁵University of Notre Dame
⁶University of Michigan
⁷University of Minnesota
⁸Wayne State University

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Abstract

Molecular simulation has emerged as an important sub-field of chemical engineering, due in no small part to the leadership of Keith Gubbins. A characteristic of the chemical engineering molecular simulation community is the commitment to freely share simulation codes and other key software components required to perform a molecular simulation under open-source licenses and distribution on public repositories such as GitHub. Here we provide an overview of open-source molecular modeling software in Chemical Engineering, with focus on the Molecular Simulation Design Framework (MoSDeF). MoSDeF is an open-source Python software stack that enables facile use of multiple open-source molecular simulation engines, while at the same time ensuring maximum reproducibility.

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Affiliations

1. Department of Chemical and Biomolecular Engineering and Multiscale Modeling and Simulation Center, Vanderbilt University, Nashville, TN
2. Department of Chemistry, Vanderbilt University, Nashville, TN
Abstract

Molecular simulation has emerged as an important sub-field of chemical engineering, due in no small part to the leadership of Keith Gubbins. A characteristic of the chemical engineering molecular simulation community is the commitment to freely share simulation codes and other key software components required to perform a molecular simulation under open-source licenses and distribution on public repositories such as GitHub. Here we provide an overview of open-source molecular modeling software in Chemical Engineering, with focus on the Molecular Simulation Design Framework (MoSDeF). MoSDeF is an open-source Python software stack that enables facile use of multiple open-source molecular simulation engines, while at the same time ensuring maximum reproducibility.

Body of Manuscript

1. Background

Molecular simulation is a methodology for predicting the collective (in particular, thermodynamic and transport) properties of systems from information about how the molecules in the system interact with each other.
That “information” can be obtained on-the-fly from quantum mechanics but, in most molecular simulations, it is encoded in a mathematical function, called a force field, that attempts to include all the intermolecular interactions between molecules (electrostatic interactions, van der Waals repulsive and attractive interactions) as well as intramolecular interactions (e.g., bond stretching, bond angle bending, and torsional interactions). More specifically, a force field is a representation of the total potential energy associated with interactions of all the atoms in the system (obtained by summing over all the molecules), which can be differentiated with respect to the position of an atom to obtain the force exerted on that atom. Force fields can be derived from first principles calculations (e.g., quantum chemistry calculations) and/or experimental data; thus, generally force fields are semi-empirical. For inhomogenous systems (e.g., a fluid adsorbed on a surface or into a pore), the force field includes models for how the molecules interact with atoms in the surfaces or with an external field. Assuming that the molecular simulation runs long enough to attain equilibrium, and that the system is large enough or configured to eliminate unwanted surface effects (through so-called periodic boundary conditions), for a given force field, molecular simulation can provide essentially exact information about the properties of the system, obtained by averaging over the configurations generated in the simulation. Two major types of molecular simulations are routinely performed: molecular dynamics (MD), in which Newton’s equations, or a convenient variation thereof, are solved for the dynamics of each atom in the system, and Monte Carlo (MC) simulation, in which configurations of the system are generated via a Markov chain process that asymptotically are distributed according to the appropriate equilibrium ensemble probability (e.g., for systems at constant molecule number \(N\), volume \(V\), and temperature \(T\), the Boltzmann distribution, in which configurations have probability \(\propto e^{-E/\beta}\), where \(E\) is the energy of the system and \(\beta = k_B T\) is Boltzmann’s constant). In either case, the raw output of the simulation is configurations of the system (known as a trajectory) that can then be analyzed to compute properties. From an MD simulation, the trajectory will consist of positions and velocities for all atoms in the system over the course of the simulation; a typical MD simulation will employ a time step of \(10^{-15}\) s, so that a 10-100 ns trajectory covers \(10^7 - 10^8\) steps. For a 100,000-atom simulation (a typical system size with current computational resources), a trajectory file can be of the order of terabytes, so that statistical analysis of such files can be thought of as a particular kind of “big data” problem.

Molecular simulation began in the 1950s with simple systems such as hard spheres (MC\(^1\) and MD\(^2,3\)) and in the 1960s with the Lennard-Jones fluid (MC\(^4\) and MD\(^5\)). For such monatomic systems, the force field is very simple, specifying the interaction energy between spherically symmetric molecules. Beginning in the 1970s, molecular simulation was introduced to the field of chemical engineering primarily by Keith Gubbins, the honoree of this Founders issue of AIChE Journal. Keith is known and admired internationally and across many disciplines not only for his contributions in molecular theory (which have been seminal, such as Gray-Gubbins perturbation theory and the statistical associating fluid theory, or SAFT, equation of state) but also for his research in molecular simulation. One of the earliest Gubbins simulation papers\(^6\) from 1979 has been cited almost 1000 times. As an aside, his postdoctoral trainee co-author on this paper, Dominic Tildesley, was for many years a successful academic in the UK before joining Unilever, where he established one of the world’s premier industrial molecular modeling groups, eventually rising to Vice President of Discovery Platforms; Tildesley also co-authored one of the seminal text books on molecular simulation.\(^7\) Keith’s influence on the field of chemical engineering in relation to molecular simulation can be measured in the Computational Molecular Science and Engineering Forum, Area 21 – is largely focused on molecular simulation) and in papers presented at Properties and Phase Equilibria for Process and Product Design conference series established in 1977 (in which the first molecular simulation paper was presented in 1980, and by 2007 more than half the presentations involved molecular simulation and/or molecular theory). Since its early days, molecular simulation has become a workhorse in science and industry. The promise of being able to predict collective properties from molecular interactions, and the attendant insight gained, have made molecular simulation (both MD and MC) an ideal and indispensable capability in materials science, biology, medicine (specifically, drug discovery) and engineering. There are commercial entities that market molecular simulation software (e.g., BIOVIA and Schrödinger). A 2002 international comparative study on molecular modeling (of which molecular simulation
constitutes a major component) documented the widespread use of molecular modeling in industry, including many chemical, drug, and personal care product companies.

The authors of this Perspective article are all beneficiaries of the trail-blazing efforts of Keith Gubbins in establishing molecular simulation as an accepted and respected subfield of chemical engineering. Today, molecular simulation is taught in most chemical engineering departments in the U.S. at the graduate level, and is increasingly available as an elective at the undergraduate level or even offered as a first-year seminar to incoming undergraduate students. It has become one of the major focuses of the educational foundation, CACHe (Computer Aids for Chemical Engineering Education, cache.org), which established a molecular modeling task force in 1998. CACHE runs a highly successful technical conference, Foundations of Molecular Modeling and Simulation (fomms.org, held every three years since 2000) that has produced many educational resources to enable chemical engineers to teach and utilize molecular simulation in the classroom. In 2012, Keith Gubbins was awarded the FOMMS Medal for his numerous and long-standing contributions to the molecular simulation community. In addition to prodigious research contributions, he has authored seminal textbooks, including the two-volume definitive treatise on the theory of molecular fluids that is an essential part of the library of any serious statistical mechanician interested in molecular fluids.

2. Development of molecular simulation tools in the chemical engineering community

Although molecular simulation (MD and MC) transcends disciplinary boundaries as noted above, chemical engineers have been particularly active in developing algorithms that compute properties of strong interest to the chemical engineering community (ChEC). One example is vapor-liquid phase equilibria, which is of enduring interest to the ChEC due to separation processes. Thus, a molecular simulation methodology for computing phase equilibrium directly and efficiently, the Gibbs ensemble MC (GEMC) algorithm, was developed in 1987 within the ChEC by Panagiotopoulos. Phase equilibria can involve differences in densities between phases of several orders of magnitude; likewise, in chemical manufacturing there can be wide ranges of state conditions. Hence, along with the development of algorithms, the ChEC has also been at the forefront of developing force fields that are accurate over wide ranges of state conditions, such as the TraPPE family of force fields optimized for vapor-liquid equilibrium (see the extensive resources at http://trappe.oit.umn.edu) and the Gaussian charge polarizable model (GCPM) for water that correctly predicts water’s phase equilibria, thermodynamic, transport and dielectric properties over wide ranges of temperature and pressure. By contrast, much of the molecular simulation community in other disciplines is focused on properties at or near ambient conditions (including ambient conditions for biological systems).

Simulation Codes

<table>
<thead>
<tr>
<th>Code</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>HOOMD-blue (Hierarchical Object Oriented MD)</td>
<td>First MD code specifically written for GPUs.</td>
</tr>
<tr>
<td>Cassandra</td>
<td>MC code with focus on complex molecules.</td>
</tr>
<tr>
<td>RASPA</td>
<td>MC/MD code for simulating adsorption.</td>
</tr>
<tr>
<td>GOMC (GPU-optimized MC)</td>
<td>GPU-accelerated general purpose MC.</td>
</tr>
<tr>
<td>MCCCS-Towhee MCCCS-MN</td>
<td>MC for complex chemical systems.</td>
</tr>
<tr>
<td>Etonica</td>
<td>Java-based MC/MD for use in education.</td>
</tr>
<tr>
<td>Other Utilities</td>
<td>Other Utilities</td>
</tr>
<tr>
<td>iRASPA</td>
<td>GPU-accelerated visualization package.</td>
</tr>
<tr>
<td>freud</td>
<td>Library for analyzing MD/MD simulations.</td>
</tr>
<tr>
<td>signac/signac-flow</td>
<td>Framework to manage and scale large simulations.</td>
</tr>
<tr>
<td>MoSDeF (Molecular Simulation Design Framework)</td>
<td>Python-based software package for molecular design.</td>
</tr>
<tr>
<td>COSMO-SAC (COnductor-like Screening Model-Segment Activity Coefficient)</td>
<td>Package to predict activity coefficients.</td>
</tr>
<tr>
<td>pysimm (python simulation interface for molecular modeling)</td>
<td>Python package for facilitating biophysical simulations.</td>
</tr>
<tr>
<td>pyprism (python-based Polymer Reference Interaction Site Model (PRISM))</td>
<td>Python-based, open-source framework.</td>
</tr>
</tbody>
</table>
Table 1. Examples of open-source molecular simulation codes and related supporting utilities developed within the chemical engineering molecular modeling community. Website links are to home pages of the codes or to code repositories. In addition, several other open-source codes emerging from the chemical engineering community are highlighted.

In this regard, the molecular simulation community in chemical engineering is particularly noted for sharing methods and capabilities by making software developed within the community freely available under open-source licenses, as described in a recent review article. Table 1 provides examples of open-source molecular simulation tools developed within the ChEC, divided into simulation codes and other utilities. Similar to GEMC, many of these algorithms developed are primarily implemented within MC and hence it is not surprising that the bulk of open-source simulation engines developed within the ChEC (see Table 1) are for performing MC simulations. The need for community-developed simulation engines, whether they are MD or MC, stems from the fact that such codes have become increasingly difficult to develop, extend, and maintain for a single individual or single research group. This is due not only to an ever growing set of features and algorithms, but also due to changes in computing hardware utilized in a research environment: we have been through the era of vector architectures (e.g., Cray, Hitachi), parallel vector computers (a small number of coupled vector processors, such as Cray YMP), massively parallel shared memory computers (MPP, such as the Intel Paragon, in which a large number of the same commodity central processor units – CPUs – used in deskside computers are linked together and communicate over a communication network), multicore processors (such as Intel Xeon that has gone from 6 cores to more than 50) both stand alone and as part of an MPP, and more recently the inclusion of massively multicore graphical processing units (GPUs, which have migrated from the gaming industry into scientific computing and data manipulation). A modern supercomputer typically consists of nodes, connected via an interconnect (from vendors such as Mellanox and Intel) These interconnects can vary from standardized ethernet connections to more specialized, proprietary high performance interconnects from various vendors. At the time of writing, the current top 500 list includes numerous systems with propriety interconnects such as Mellanox Infiniband (now owned by Nvidia), Intel Omni-Path, Cray Aries, and Fujitsu Tofu along with standard ethernet connections ranging from 10G to 100G, where each node houses multiple commodity multicore CPUs and GPUs. This is the dominant architecture of the supercomputers on the top 500 list of the fastest computers in the world, with the top 5 supercomputers having between 1.5 and in excess of 10.5 million total computing cores at the time of writing; designing and maintaining simulation codes that perform efficiently on these rapidly evolving computer architectures is a significant challenge. Beyond community developed simulation engines, we have also seen the rise of other community developed utilities to support simulation, e.g., in the form of general analysis packages as well as software that makes it easier to accurately and reproducibly initialize configurations, apply force fields to molecules, and create input files for a variety of simulation engines.

In the remainder of this Perspective, as an example of ChEC open-source software, we focus our discussion on the Molecular Simulation Design Framework (MoSDeF), to which all the authors are contributors. MoSDeF is a set of Python tools to facilitate the initialization and parameterization of systems, with the goal of enabling transparent and reproducible molecular simulation workflows that, at the same time, are user-friendly and extensible.
3. Molecular Simulation Design Framework (MoSDeF)

As shown in Figure 1, performing a molecular simulation, whether MD or MC, requires multiple steps: building an initial configuration of the system, selecting and applying a force field, generating a syntactically correct input file (or files) for a target simulation engine, equilibration (to relax the system from its initial configuration – e.g., a crystal – to a configuration characteristic of equilibrium – e.g., liquid), production run to generate a trajectory, and analysis of the trajectory (e.g., averaging over the trajectory to compute thermodynamic and/or structural properties, perform visualization, etc.). Often reliability and statistics are improved by running multiple independent trajectories using the same workflow. Accomplishing these steps in a way that is both accurate and reproducible can be a significant challenge. For example, the application of a force field is a frequent source of error in simulations; for a system composed of moderately complex molecules (such as an ionic liquid) the force field can have a hundred or more parameters that must be provided, offering multiple opportunities for errors (e.g., use of incompatible units, use of parameter values from a publication containing a typographical error, incorrect application of parameters due to logic errors or because of ambiguous definition of parameter usage, etc.). While the use of a community developed, open-source simulation engine may help to reduce the likelihood of fundamental errors in algorithms underlying the simulations, such codes cannot necessarily prevent users from providing parameters that are inconsistent with the intended usage.

Typically, many of these steps are performed within a given research group by a single graduate student, often making use of ad hoc, in-house software, even if open-source simulation engines are used. This approach has several shortcomings that can make simulations more prone to error, limit the extensibility, and hamper reproducibility. For example, the various tools used to accomplish these steps may only be loosely coupled and require manipulation, editing, and/or modification of the tools and/or data by the user. This manipulation may introduce errors and make it difficult to reproducibly capture the exact procedures employed. The need for human manipulation may also limit the ability to use such workflows in applications that require automation, such as parameter screening studies or within the context of larger workflows (e.g., to predict phase equilibrium within a process simulator). The use of in-house software itself, which is typically not
open-source or freely available, creates numerous roadblocks as well. Someone wishing to reproduce a simulation would be required to write their own software to accomplish the same tasks. The development of such software may be time consuming and publications often do not provide sufficient detail regarding the procedures used to initialize and parameterize simulations. Furthermore, without access to the original source code, it is not possible to ascertain the quality of the software; that is, to know whether it has undergone sufficient validation or if there are errors and bugs that ultimately impact the accuracy of the reported results.

The Molecular Simulation Design Framework (MoSDeF)\textsuperscript{32} is designed to address these issues of automation/efficiency, accuracy, and reproducibility in molecular simulation. MoSDeF is an open-source Python library built upon the scientific Python software stack with three major components: mBuild (for constructing initial configurations of systems) and foyer (for applying force fields). The third component, GMSO (General Molecular Simulation Object), is currently under development and is designed to be a general, flexible way of encapsulating the information required to define a simulation topology in a simulation engine in an agnostic manner. All of the capabilities of MoSDeF are scriptable, thus making the tools inherently reproducible, as well as suitable for automated calculations (e.g., screening). MoSDeF is implemented as a set of composable/modular tools, where each “subpackage” (i.e., module) is designed such that it can be used within MoSDeF, or as a standalone package, allowing MoSDeF to more easily integrate with other community efforts. This also allows the framework to be more easily modified, tested, extended, and have fewer bugs than a monolithic approach. Performing a simulation using MoSDeF, combined with dissemination of simulation scripts on a service such as Github, enables a molecular simulation to be published as a TRUE (t transparent, r reproducible, u usable by others, and e extensible) simulation\textsuperscript{33}.

MoSDeF has its origins in a decade of National Science Foundation (NSF)-supported collaborative research at Vanderbilt University involving researchers from chemical engineering and computer science\textsuperscript{34–36}, the latter affiliated with the Institute for Software Integrated Systems (ISIS)\textsuperscript{37}. ISIS is a leading academic software engineering research center, and is the originator of the concept of model-integrated computing (MIC)\textsuperscript{38}. MIC is a systems engineering approach that focuses on the creation of domain specific modeling languages to capture the essential features of the individual components of a given process, at the level of abstraction that is appropriate for the end users. Due to abstraction, processes are described at a meta level that allows tasks to be coupled together to execute scientific or engineering workflows. MIC has been deployed in applications as diverse as managing auto assembly lines and processing health records. MIC design principles, domain-specific modeling languages, and the general philosophy of abstraction have shaped the development of MoSDeF. In particular, MoSDeF attempts to be simulation-engine-agnostic, treating the concept of a molecular simulation at a meta level, above the specifics of the simulation engines. The tools within MoSDeF are designed to fully describe a system: implementation relies on writers to instantiate syntactically correct input files for specific engines from this information. MoSDeF was initially developed to support several commonly used open-source MD codes (LAMMPS\textsuperscript{39}, GROMACS\textsuperscript{40} and HOOMD-blue\textsuperscript{41}) and has since grown to support open-source MC simulation engines, namely Cassandra\textsuperscript{16} and GOMC\textsuperscript{18}.

In the Supplementary Information, we provide details on how to install MoSDeF through various hosting systems (anaconda, docker, from source using github, etc.) on Apple OSX, Linux, and Windows platforms. Below we describe each of the three key components. Source code, tutorials, documentation, and related publications can be accessed from mosdef.org and/or github.com/mosdef-hub/.

3.1. mBuild

Hosted file


As shown in Figure 1, the first step in a simulation workflow typically involves defining the configuration of the atoms (or more generally, particles) in the system. The mBuild Python library\textsuperscript{42,43} has been developed to
be a general, customizable tool for constructing arbitrarily complex system configurations in a programmatic fashion (i.e., scriptable). Key to the mBuild library is its underlying Compound data structure. A Compound is a general “container” that can describe effectively anything: an atom, a collection of atoms, a molecule, a generic point particle, a collection of Compounds, operations on the underlying Compounds and/or data, etc. Compounds can be duplicated, rotated, translated, scaled, etc. to construct a system. Compounds can also contain information regarding connections between the atoms, by defining either fixed Bonds within a Compound or by adding Ports that allow connections to be made between separate Compounds. Ports define both location and orientation of a connection; in atomistic systems, the number of Ports and their locations are typically representative of the underlying chemistry. For example, Figure 2 shows Python code that defines a CH$_2$ moiety with two C-H Bonds and two Ports. In order to create a connection between two Compounds, a user simply states which Ports should connect and mBuild automatically performs translations and reorientations, creating a new (composite) Compound (see Klein et al. for more details). As such, this allows complex systems to be built-up from smaller, interchangeable pieces that can be connected, through the use of the concept of generative modeling. This design approach allows for declaratively expressing repetitive structures, such as polymer chains and planar tilings (as used in Figure 2) and also allows significant modifications to system structure/chemistry to be made with only minimal changes to the initialization routines.

3.2. Foyer

After a system configuration is initialized, the interactions between all constituents must be defined before a system can be simulated (as shown in Figure 1), i.e., the force field must be applied to the system. The Foyer library has been developed as a general tool for applying force fields to molecular systems (i.e., atom-typing), that provides a standardized approach to defining chemical context and atom-typing rules. In Foyer, the forcefield parameters and the rules that dictate parameter usage are stored together in a standardized XML file, separate from the code used to evaluate them. Usage rules are encoded by using a combination of a SMARTS-based annotation scheme, which defines the chemical context associated with a given parameter, and overrides that define rule precedence. SMARTS is a language designed for describing molecular patterns, thus allowing information about the bonded environment of an atom to be efficiently and clearly encoded in a format that is both human and machine readable. For example, the chemical context of a terminal methyl group (-CH$_3$) in an alkane can be expressed as [C;X4](C)(H)(H)H. In this annotation, [C;X4] indicates that the atom of interest is a carbon (C), with 4 total bonds (X4) and (C)(H)(H)H provides the identity of those 4 bonds (1 carbon, 3 hydrogens). Figure 3 shows a snippet from the Foyer XML forcefield file demonstrating how these usage rules can be encoded, using select parameters from OPLS-AA force field (See Klein et al. for more details). By separating the usage rules and parameters from the software used to evaluate them, the Foyer library does not need to change if changes are made to a force field file. As such, this allows the implementation of novel and “custom” force fields without the need to write new software, which simplifies the process of disseminating and evolving forcefields, and increases reproducibility of work by making it clear not just what force field was used, but how it was applied to the system. A complimentary approach not requiring SMARTS and overrides is to make molecule-specific XML files available (e.g., via webpages such as http://trappe.oit.umn.edu).

3.3. General Molecular Simulation Object (GMSO)

With a system initialized and parameterized, the information in the system topology must be written to a file for a simulation engine. While the information required by different simulation engines is, generally speaking, the same, the structure and format of the data file(s) passed to simulation engines is typically unique to the engine itself. Generating these files accurately, especially for a wide range of unique simulation engines, can be non-trivial. The current version of MoSDeF relies upon the use open-source utilities parmed to store this information; these tools along with native MoSDeF code, include parsers to generate syntactically correct data files. In this approach, a single simulation topology can be used to
generated input files for a variety of simulation engines, allowing different engines and methodologies (e.g., MC and MD) to be applied to the same system. While effective, these backend codes do not have general support for the breadth of simulation engines and force fields we aim to include. To this end, the General Molecular Simulation Object (GMSO) has been under development with the goal of becoming the *de facto* backend data structure of the MoSDeF. The goal of GMSO is to serve as a general container for all of the relevant system information (e.g., the fully parameterized system), stored in a simulation engine agnostic way. GMSO is designed with interoperability and support for various functional forms as a first-class feature. For example, GMSO builds upon the idea of Foyer XML data file, shown in Fig. 3, but provides further meta data; this includes encoding the functional forms of the potentials in the force field (those that can be expressed in computer algebraic inputs) using the sympy Python library. GMSO is also structured to make it easier to add data file writers, allowing GMSO support to be extended and customized. Because GMSO supports user-defined analytic equations for force field components, it future-proofs GMSO for new developments in force fields, such as those being pursued by several of the authors.

3.4. Computational Screening and Automation using MoSDeF

Since all the functions of MoSDeF are scriptable, when combined with a workflow management tool such as signac/signac flow\[^{21}\], it is relatively trivial to perform computational screening of the properties of systems by looping over chemistries and/or conditions and calculating relevant properties from the simulations. The MoSDeF/signac combination has been used to screen the impact on nanolubrication properties of end-group chemistry of self-assembled alkylsilane tethers on amorphous silica surfaces\[^{23}\], leading to a machine-learning-derived model connecting end-group cheminformatic descriptors with tribological properties of interest. In another example\[^{50}\], the diffusivities of ions in organic solvents were screened for 22 different solvents, revealing a pattern in this large data set (ion diffusivity proportional to solvent diffusivity) that was in contrast with previous, primarily experimental findings (ion diffusivity proportional to solvent dipole moment). The computational screening finding were confirmed in subsequent experimental studies utilizing quasi-elastic neutron scattering\[^{51}\] and NMR\[^{52}\].

3.5 Expanding MoSDeF

As noted earlier, the genesis of MoSDeF was a series of NSF grants to Vanderbilt PIs Cummings, McCabe, Iacovella, and Ledezci\[^{34–36}\]. A recent collaborative NSF grant\[^{53}\] has funded groups from the universities of Michigan (Glotzer and Anderson), Notre Dame (Maginn), Minnesota (Siepmann), Delaware (Jayaraman), Houston (Palmer), Wayne State (Potoff), and Boise State (Jankowski) universities to work together to expand MoSDeF’s capabilities, including the collaborative design and development of the aforementioned GMSO backend. This collaboration is resulting in increasing integration with HOOMD-blue, integration with MC codes Cassandra and GOMC, and the first principles MD/MC code CP2K; additionally, MoSDeF has been integrated more closely with Michigan’s signac workflow management tools. In the case of Cassandra, for example, using MoSDeF existing utilities and adding additional capabilities resulting from the Vanderbilt/Notre Dame collaboration, the complexity of setting up a simulation has been reduced from 9 steps (including 3 requiring user editing of files) to a single python script using MoSDeF; this, in turn, has enabled computational screening with Cassandra. Other groups, including Houston, Boise State, and Delaware, are focusing on developing modules to implement complex workflows and analyses involved in phase equilibrium calculations and construction of intricate molecular models. Building the modules around the MoSDeF framework will enable these workflows to be performed in a reproducible fashion with a variety of widely used simulation engines.

An example of the capabilities enabled by this collaboration is given in the Supplementary Information (SI). Inspired by the honoree of this special issue, Keith Gubbins, in the SI we report the use of five different simulation codes (the open-source MC codes Cassandra and GOMC, the open-source MD codes LAMMPS and GROMACS, and the open-source first principles MD code CP2K) to repeat calculations reported by Striolo et al.\[^{54}\] on the adsorption of water into carbon slit pores. The latter were groundbreaking simulations for their time and the paper has been cited ~200 times (Google Scholar). The paper reported
adsorption/desorption isotherms, demonstrating the hysteresis seen in experiment, as well as density profiles and orientational structure of the adsorbed water into carbon slit pores. The Striolo et al. simulations were performed using in-house codes; thus, they are almost impossible to reproduce in detail. In the SI, we show that we can reproduce the adsorption/desorption isotherms reported by Striolo et al. to within an acceptable degree using Cassandra and GOMC; more importantly, we show that by using the MoSDeF tools to create the simulations, we can easily test multiple engines, and show we get excellent agreement between the two different MC codes. Having used the technique of GEMC in both Cassandra and GOMC, we establish the number of water molecules in the pore at a given external pressure. We then perform \textit{NVT} (constant number of molecules, volume and temperature) simulations using multiple codes. We find remarkable agreement for the water structure inside the pore between the MC engines Cassandra and GOMC and MD engines LAMMPS and GROMACS. The use of MoSDeF (mbuild to build the simulation systems and foyer to apply the force fields) is absolutely essential to obtaining consistency between these calculations. The first principles MD code CP2K with interactions described on-the-fly via Kohn-Sham density functional theory produces similar, but not identical, results for water structure, thereby allowing us to identify differences in water-substrate interactions. The fact that one can move the simulated system between all of these codes fairly effortlessly, thanks to the use of the MoSDeF tools and its meta-level abstraction of the concept of molecular simulation, is a very significant step forward for the simulation community. Moreover, the SI contains all the instructions needed for the reader to download and run all the utilities and codes needed to reproduce the reported calculations exactly, hence qualifying these as TRUE simulations.\textsuperscript{33}

4. Conclusions

For several decades, the open-software movement has been making its presence felt in the chemical engineering community. Open-source software offers many advantages over proprietary codes. First, they are universally available and do not contain any hidden parameters. This makes verification of results published using these codes much more feasible than for proprietary codes. Indeed, some scholarly journals have taken the position of considering only manuscripts for publication in which molecular modeling calculations were performed using open-source codes or source code that is made available to reviewers. Second, open-source codes are available at no cost, which means that the codes can be downloaded and used by researchers throughout the world, removing barriers for scientific progress. Third, open-source codes typically attract a community of users and/or developers, so that bugs are discovered and eliminated quickly, often overnight; in the case of proprietary software, bugs are typically only fixed during update cycles, which may be months apart, or may even go unnoticed, since the code cannot be inspected by users. The downside of open-source software is that, since there is no revenue stream in the usual sense (sale of software), the sustainability of an open-source code over decades can be questionable. However, codes can reach a level of usage such that the effort to maintain and improve the code is taken on by the user community; LAMMPS has arguably reached this position. Also, for some open-source codes there is an alternative revenue stream. For example, Red Hat is the biggest contributor and supporter of the open-source Linux operating system. It makes money by writing, selling, and supporting business-oriented middleware that runs within Linux, as well as selling consulting services to companies switching to Linux for their enterprise software. The commercial Scienomics MAPS platform for materials and process simulations embeds some of the open-source MD and MC codes, such as LAMMPS, Cassandra, and MCCCS-Towhee. Enthought, Inc. is a software company based in Austin, Texas, that develops and markets scientific and analytic computing solutions using primarily the Python programming language; its commercial activities underwrite the widely used open-source SciPy (Scientific Python) package.

We dedicate this Perspective to our colleague, mentor, and friend, Keith Gubbins. The authors of this article wish to express their deep gratitude to Keith for all he has done for our community. We wish him many more years of productive science.

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