Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique

Christi Jose¹, Aishwarya Singh¹, Kavyashree N Kalkura¹, George V Jose², Shailina Srivastava¹, Rameshchan K A³, Shweta Yadav⁴, R. Ravikrishna¹, Meinrat O. Andreae⁵, Scot T. Martin⁶, Pengfei Liu⁷, and Sachin S Gunthe⁸

¹Indian Institute of Technology Madras
²Indian Institute of Technology Bombay
³College of Engineering Munnar
⁴Central University of Jammu
⁵Max Planck Institute for Chemistry
⁶Harvard University
⁷Georgia Institute of Technology
⁸Environmental and Water Resources Engineering Division, Department of Civil Engineering, Indian Institute of Technology Madras, Chennai 600036, India.

November 22, 2023

Abstract

Comprehending the intricate interplay between atmospheric aerosols and water vapour in subsaturated regions is vital for accurate modelling of aerosol-cloud-radiation-climate dynamics. But the microphysical mechanisms governing these interactions with ambient aerosols remain inadequately understood. Here we report results from high-altitude, relatively pristine site in Western-Ghats of India during monsoon, serving as a baseline for climate processes in one of the world’s most polluted regions. Utilizing a novel quartz crystal microbalance (QCM) approach, we conducted size-resolved sampling to analyse humidity-dependent growth factors, hygroscopicity, deliquescence behaviour, and aerosol liquid water content (ALWC). Fine-mode aerosols (<2.5 μm) exhibited size-dependent interactions with water vapour, contributing significantly to ALWC. Deliquescence was observed in larger aerosols (>180 nm), influenced by organic species, with deliquescence relative humidity (DRH) lower than that of pure inorganic salts. This research highlights the significance of understanding ambient aerosol-water interactions and hygroscopicity for refining climate models in subsaturated conditions.
Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique

Christi Jose¹², Aishwarya Singh¹², Kavyashree N. Kalkura¹², George V. Jose³, Shailina Srivastava¹², Rameshchand K. A.⁴, Shweta Yadav⁵, R. Ravikrishna²⁶, M. O. Andreae⁷⁸⁹, Scot T. Martin¹⁰¹¹, Pengfei Liu¹²*, and Sachin S. Gunthe¹²,*

¹Environmental Engineering Division, Dept of Civil Engineering, Indian Institute of Technology Madras, Chennai 600036, India
²Centre for Atmospheric and Climate Sciences, Indian Institute of Technology Madras, Chennai 600036, India
³Dept of Civil Engineering, Indian Institute of Technology Bombay, Mumbai 400076, India
⁴Dept of Mechanical Engineering, College of Engineering Munnar, Munnar 685612, India
⁵Dept of Environmental Sciences, Central University of Jammu, Samba, Jammu and Kashmir 181143, India
⁶Dept of Chemical Engineering, Indian Institute of Technology Madras, Chennai 600036, India
⁷Multiphase Chemistry Department, Max Planck Institute for Chemistry, 55128 Mainz, Germany
⁸Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA 92093, USA
⁹Department of Geology and Geophysics, King Saud University, Riyadh 11451, Saudi Arabia
¹⁰Department of Earth and Planetary Sciences, Harvard University, Cambridge, MA 02138, USA
¹¹John A. Paulson School of Engineering & Applied Sciences, Harvard University, Cambridge, MA 02138, USA
¹²School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA 30332, USA

*Correspondence to: Sachin S. Gunthe (s.gunthe@iitm.ac.in) and Pengfei Liu (pengfei.liu@eas.gatech.edu)

KEYPOINTS

- QCM technique revealed aerosol properties at low RH values offering insights that may not be captured by traditional size-based measurements
- The size-dependent variations in aerosol properties below DRH emphasize the need to understand how aerosol properties change with RH history
- Organic in aerosols complicate its cloud forming ability, necessitating additional studies in diverse environments to improve climate models
ABSTRACT

Comprehending the intricate interplay between atmospheric aerosols and water vapour in subsaturated regions is vital for accurate modelling of aerosol–cloud–radiation–climate dynamics. But the microphysical mechanisms governing these interactions with ambient aerosols remain inadequately understood. Here we report results from high-altitude, relatively pristine site in Western-Ghats of India during monsoon, serving as a baseline for climate processes in one of the world’s most polluted regions. Utilizing a novel quartz crystal microbalance (QCM) approach, we conducted size-resolved sampling to analyse humidity-dependent growth factors, hygroscopicity, deliquescence behaviour, and aerosol liquid water content (ALWC). Fine-mode aerosols (<2.5 µm) exhibited size-dependent interactions with water vapour, contributing significantly to ALWC. Deliquescence was observed in larger aerosols (>180 nm), influenced by organic species, with deliquescence relative humidity (DRH) lower than that of pure inorganic salts. This research highlights the significance of understanding ambient aerosol-water interactions and hygroscopicity for refining climate models in subsaturated conditions.

PLAIN LANGUAGE SUMMARY

Aerosol particles interact with water vapour in the atmosphere. Understanding these interactions in sub – and super-saturated regions is crucial because they affect processes such as cloud formation, radiation, and climate. We collected ambient aerosol samples from Western Ghat in India to understand natural processes that happen without significant human interventions. Using a special technique involving a quartz crystal sensor, we measured different aspects, such as how aerosols grow with humidity, their ability to uptake the water molecules, and when they transition from solid to liquid (deliquescence). Interestingly, fine particles, especially those smaller than 2.5 µm, exhibited unique behaviours. While larger particles underwent a transition from solid to liquid under certain conditions, this didn’t happen for the smaller particles. These findings highlight the importance of understanding these interactions for more accurate climate predictions.
1. INTRODUCTION

Understanding the hygroscopicity of atmospheric aerosols is crucial for assessing cloud formation and their climate and air pollution impact (Cheung et al., 2015). Investigation of the hygroscopicity of ambient aerosols poses significant challenges due to their chemical complexity, particle size variation, phase state, and viscosity. Techniques like Fourier transform infrared spectroscopy (FTIR) (Y. Liu et al., 2008; Y. Liu & Laskin, 2009), quartz crystal microbalance (QCM) (Chao et al., 2020; Demou et al., 2003; P. Liu et al., 2016a; P. Liu, Song, et al., 2018a), Raman spectroscopy (Ling & Chan, 2008; Y. J. Liu et al., 2008), electrodynamic balance (EDB) (Choi & Chan, 2002; Peng & Chan, 2001; Pope et al., 2010), optical microscopy (OM) (Ahn et al., 2010; Eom et al., 2014; Gupta et al., 2015), hygroscopicity tandem differential mobility analysis (HTDMA) (Cheung et al., 2015; Prenni et al., 2007; Zieger et al., 2017a) and size-selected cloud condensation nuclei (CCN) spectrometry (Petters et al., 2007; Pöhler et al., 2016; Rose et al., 2008) have been employed to study the hygroscopicity of laboratory–generated and ambient aerosols (Tang et al., 2019). However, a consensus on the most effective method is lacking, leading to inconsistent and incomparable results. Hygroscopicity measurements primarily focus on accumulation mode particles, limiting data on nucleation and coarse mode particles with complex behaviour. Discrepancies in hygroscopic properties and organic and inorganic species composition in atmospheric aerosols result in diverse growth factors and phase transitions, crucial for accurate climate modelling (Li et al., 2021).

The QCM is effective in determining mass–based hygroscopicity and physical property variations of atmospherically–relevant aerosols (Demou et al., 2003; P. Liu et al., 2016b; P. Liu, Song, et al., 2018b). It overcomes the limitations of traditional methods like HTDMA, which has a narrow particle size range with limited relative humidity (RH) resolution (Zhao et al., 2022). Conventional HTDMA methods inadequately reveal hygroscopicity heterogeneity and phase transitions in wider aerosol size range, impacting processes like CCN activation and aerosol liquid water (Li et al., 2021). QCM’s mass-based measurements accurately quantify aerosol hygroscopicity behaviour across a wide RH range in the atmosphere, detailing physical property variations during phase transitions and help to delineate water uptake mechanisms, including solubility and water diffusion limitations (P. Liu, Li, et al., 2018; P. Liu, Song, et al., 2018b).
India's climate is distinct and intricate, and as per the 2020 climate change assessment report, India has faced challenges like temperature rise and extreme weather events since the mid-20th century. A key hurdle in addressing these issues is the absence of systematic aerosol characteristic measurements, particularly in aerosol-water vapor interactions in the subsaturated regime (Cheung et al., 2015). Aerosol particles from a forest-surrounded by pristine area are anticipated to contain substantial biogenic organic aerosols. The dynamic chemical transformation of secondary organic aerosols (SOA) in the atmosphere leads to intricate compositions. Atmospheric processing further complicates the understanding of phase state and properties of atmospheric aerosols (Rastak et al., 2017). In this study, QCM measurements are utilized to explore aerosol hygroscopicity dependence on RH, chemical composition, and size across a broad range of ambient aerosols in India’s relatively pristine setting.

2. MATERIALS AND METHODS

Size-resolved ambient aerosols were sampled at the Natural Aerosol and Bioaerosol High Altitude (NABHA) Laboratory at the College of Engineering, Munnar (10.0930° N, 77.0682°E; 1600 m above mean sea level), a high-altitude and typically clean site situated in the Western Ghats of India, using a 10-stage micro-orifice uniform deposit impactor (MOUDI-II 120-R, TSI) (V. Marple et al., 2014; V. A. Marple et al., 1991) during the monsoon season (August-September) of 2021. Ambient particles collected on PTFE (Polytetrafluoroethylene) filters were transferred to a hydrophobic SiO$_2$-coated quartz sensor by gently pressing the filter paper onto the sensor. The hygroscopic growth factor, hygroscopicity parameter and deliquescence relative humidity (DRH) were estimated using a highly sensitive mass balance instrument, QCM (QSense Analyzer, Biolin Scientific) (P. Liu, Song, et al., 2018a; Reviakine et al., 2011). The accuracy and robustness of the method were demonstrated by performing measurements of the hygroscopic growth factor and DRH for sucrose and (NH$_4$)$_2$SO$_4$ particles, respectively, which were compared with previously reported results (Arenas et al., 2012a; Chao et al., 2020; Martin, 2000; Norrish, 1966; Peng et al., 2022; Starzak & Peacock, 1997; Zobrist et al., 2011) (Fig. S3 and S4). The hygroscopic growth factor and the corresponding hygroscopicity parameter, $\kappa$, were determined over a wide range of RH conditions using $\kappa$-Köhler theory (Petters & Kreidenweis, 2007), where $\kappa$ represents a quantitative measure of aerosol water uptake characteristics and CCN activity (Zhao et al., 2022). The water uptake characteristics and the phase transition behaviours of the ambient
samples were investigated and compared between different size ranges of particles below 10 μm collected using the MOUDI sampler. The Aerosol Liquid Water Content (ALWC) (Bian et al., 2014) at different RH conditions was also estimated from corresponding hygroscopicity parameters for each size range of particles (Fig. S5). The number size distributions of the ambient aerosol particles of size 10 to 420 nm [measured in parallel using a scanning mobility particle sizer – SMPS, comprised of an electrostatic classifier (EC, TSI Model 3082) equipped with a differential mobility analyser (DMA, TSI Model 3081) and a condensation particle counter (CPC, TSI Model 3750)], were then converted to mass size distributions assuming a particle density of 1.2 g cm$^{-3}$ (DeCarlo et al., 2004). The supermicron particle mass size distribution for the same season, but a different year, was used in this study, which had been obtained using an Ultraviolet Aerodynamic Particle Sizer (UV-APS, TSI Inc., model 3314) (Valsan et al., 2016).

Meteorological parameters were recorded using an automatic weather station (AWS, Clima Sensor US) during the sampling period. Figure 1 shows the sampling site – Munnar, along with aerosol optical depth (AOD) over India during the monsoon season, indicating the relatively pristine nature of the sampling site. During the measurement, airmasses predominantly arrived from the southwest direction, bringing clean marine influx to the observational site and resulting in a low influence of anthropogenic emissions (Fig. 1). A more detailed description of the instruments, experimental techniques, and estimation of various parameters is provided in the Supporting Information (SI).
Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon season of 2021 (June - September). The AOD distribution clearly indicates a relatively low aerosol loading over the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind rose diagram shown in the inset shows the average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August – September 2021). The prevailing air masses mostly originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the observational site, confirming the relatively low influence of anthropogenic activities.

3. RESULTS AND DISCUSSION

The QCM sensor’s oscillation frequency variation ($\Delta f$) signifies water uptake and release, providing insights into adsorption, desorption, and physical states of particles during solid-to-aqueous phase transition (Arenas et al., 2012b). Normalizing the frequency shift at higher RH to that of deposited dry sample (RH <5%) yields the percentage value, $\Delta f_N$ (Chao et al., 2020). Figure 2 shows $\Delta f_N$ for the ambient aerosol particles across various RH levels and sampled size ranges. A negative sign in $\Delta f_N$ indicates frequency reduction with increasing RH due to water uptake. Further details of $\Delta f_N$ and its derivative, $d(\Delta f_N)/(\text{RH})$, are available in SI.

For the particle size <180 nm (Fig. 2a), $\Delta f_N$ decreased, implying water uptake, with <56 nm particles exhibiting the lowest value. In Figure 2b and 2c, particles showed increased water uptake at lower RH, evidenced by $\Delta f_N$ decrease. However, $\Delta f_N$ suddenly increased at specific RH, indicating a drastic water uptake leading to deliquescence-induced phase transition. Figures 2a-c revealed deliquescence in particles >180 nm at different RH (known as DRH), absent in those <180 nm. Past studies reported no deliquescence for ambient and laboratory-generated atmospherically relevant particles even at high RH, attributing them to organic nature (Arenas et al., 2012a; Brooks et al., 2002; Chao et al., 2020; Peng et al., 2022). We hypothesize that particles <180 nm were likely dominated by freshly formed secondary organic aerosols (SOA) from biogenic volatile organic compound (VOC) oxidation, supported by the densely vegetated/forest region and season. Concurrent quasi-continuous measurements during the same campaign indicated upto ~90% organic fraction in NR-PM$_1$ (non-refractory particulate matter with an aerodynamic diameter ≤1 μm; Tab. S2) with details discussed in subsequent studies.
Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For panels a, b, and c, $\Delta f_s$ represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) conditions normalized to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in the value of $\Delta f_s$ for each size range for the sampled ambient aerosol particles indicates the water uptake at different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size ranges. In panels d, e, and f, the derivative of $\Delta f_s$ with respect to RH ($d(\Delta f_s)/d(RH)$) is plotted against RH to determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. The RH values at which $d(\Delta f_s)/d(RH)$ becomes ≥1 (marked by the dotted line) represent the DRH values for the individual aerosol size ranges.

Following Chao et al. (2020), DRH, indicated by $d(\Delta f_s)/(RH)\geq 1$ ranged from ~60 – 68% for three size ranges (180 nm – 320 nm; 320 nm – 560 nm; 560 nm – 1 µm) in this study (Fig. 2e). Above 180 nm, inorganic salts, like (NH$_4$)$_2$SO$_4$, may contribute to DRH appearance (Hu et al., 2010) (Fig. 2b,e). These DRH values, smaller than pure (NH$_4$)$_2$SO$_4$ (DRH=80%), suggest mixed salts and/or organic compounds in the sampled aerosol particles. The deliquescence behaviour of organic-inorganic mixtures introduces complexity due to solubility limitations, influenced by the organic species in the aerosol particles. Previous studies noted reduced DRH of inorganic species and sometimes, unaffected by organics (Smith et al., 2011, 2012, 2013). Thus, we hypothesize 180 nm to 1 µm particles at this site reflect a complex mixture of inorganic salts like (NH$_4$)$_2$SO$_4$ and water-soluble organic compounds. Supermicron particles (>1 µm), except 1 – 1.8 µm, showed <17% $\Delta f_s$, indicating lower water uptake (Fig. 2c). Beyond maximum water uptake, the three size ranges exhibited different DRH values. 1.8 – 3.2 µm and 5.6 – 10 µm exhibited DRH between 56 – 68%, suggesting a mixture of water-soluble organic compounds and inorganic salts. However, 3.2 – 5.6 µm showed a very high DRH (>81%) with the organic fraction minimally affecting DRH due to solubility limitations. For 1.8 – 3.2 µm and 5.6 – 10 µm, reduced DRH
compared to pure inorganic salts suggests a mix of water-soluble organic and inorganic compounds (Smith et al., 2011, 2012, 2013). The 1.0 – 1.8 µm range exhibited a >30% ΔfN reduction at ~42% RH, indicating a higher water uptake and the observed DRH appeared to be ~45 – 53%, consistent with the highest κm (~0.44, see discussion below) among all measured size ranges (Chao et al., 2020). The low DRH at this site suggests presence of highly water-soluble organic material within the organic-inorganic (sea salt) mixture in this size range.

Figure 3 shows the size-resolved mass-based hygroscopic growth factors (gf_m) and the corresponding mass-based hygroscopicity parameters (κ_m) of ambient aerosol particles calculated from QCM measurements (Text S4 and S5) over a wide RH range (2 – 90%). In Figure 3a, the growth factor increases with water uptake in humid air, showing size-dependent variations across different RH levels (indicated by markers). Each size range had a maximum RH limit beyond which gf_m estimation was impossible due to deliquescence, as observed in previous studies (Chao et al., 2020). Accordingly, the highest gf_m=1.72 was recorded at RH=90% for the 100 – 180 nm particles. The pronounced variations in gf_m between different-sized particles at a given RH indicate varying chemical composition and hygroscopicity across the sampled sizes.

Figure 3b shows the size-dependent average κ_m values, calculated from the gf_m values in Figure 3a. Each particle size range exhibited minimal κ_m variation (average κ_m ± 0.007), suggesting ideal solution behaviour across the wide RH range (Z. Wang et al., 2017). The average κ_m values (indicated by dashed lines in Figure 3b) of the size-resolved ambient particles varied significantly between 0.016 (<0.056 nm) and 0.44 (1.0 – 1.8 µm) over the investigated RH range. This implies κ_m as a function of size, and indicates distinct chemical compositions for each size fraction. For example, the small increase in κ_m from 0.016 to 0.078 for particles <56 nm to 56-100 nm, respectively, is potentially owing to aging processes (Zhang et al., 2023). Aerosol particles <320 nm exhibited very low average κ_m (0.06), which may be primarily due to the strong dominance of organic compounds in these size ranges (Demou et al., 2003; Li et al., 2021; Shi et al., 2022). The particles in the size range of 320 nm – 1 µm were moderately hygroscopic (average κ_m=0.19) indicating the influence of inorganic salts (Wu et al., 2016). The particles between 1.0 – 1.8 µm exhibited highest κ_m (0.44), which implies the presence of highly hygroscopic material such as sea salt.
(Zieger et al., 2017b) mixed with organic compounds. Even larger particles exhibited reduced κm, possibly due to the presence of dust particles (Koehler et al., 2009), transported at this site during monsoon season (Valsan et al., 2016). The κ-Köhler parameterization fit accurately elucidates the gf_m within each size range over a wide range of RH values (Petters & Kreidenweis, 2007).

Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor (gf_m) derived using a quartz crystal microbalance (QCM) for ten different size ranges of ambient aerosol particles at different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the corresponding κ-Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity parameters, κm. The error bars represent the variations in gf_m averaged over the mass change corresponding to different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the mean κm values corresponding to the respective size ranges. (b) The data points are the κm values calculated based on the gf_m (as shown in (a)) using the κ-Köhler theory for different RH conditions in the subsaturated regime. The dotted lines represent the mean κm value for each size range of ambient aerosol particles and the error bars represent one standard deviation.

Figure 4 shows the size dependency of κm in the subsaturated regime measured by QCM together with κCCN (Text S2) in the supersaturated regime determined by size-resolved CCN measurements and aerosol size distributions obtained by SMPS and UV-APS. The submicron particles show a bimodal number size distribution with an Aitken mode peak at 63 nm and an accumulation mode at 145 nm, while the mass size distributions peak at 350 nm and 3 μm,
respectively. The average $\kappa_{\text{CCN}}$ increased from a value of 0.28 for the Aitken mode particles to 0.47 for the accumulation mode particles. In the submicron region, $\kappa_m$ exhibited a nominal increase with particle size (Wu et al., 2016), in parallel with the increase in $\kappa_{\text{CCN}}$, albeit at much lower absolute values. This increase likely reflects the decrease of organic fraction with size in the submicron range, which is supported by the appearance of deliquescence in the size range above 180 nm, indicating the presence of inorganic salts. In the supermicron region (for particles $\geq 1 \mu m$), $\kappa_m$ peaked due to the potential presence of NaCl particles in the size range 1.0 – 1.8 $\mu m$, and decreased again in the higher size ranges ($>1.8 \mu m$) likely because of the presence of dust particles. The average $\kappa_m$ value (0.18) obtained from QCM measurements across all the sampled size ranges in the subsaturated regime was lower than the average $\kappa_{\text{CCN}}$ (0.39), likely due to the solubility limitation (Dusek et al., 2011; Hersey et al., 2013; Rastak et al., 2017; Riipinen et al., 2015; Wittbom et al., 2018) of organic and inorganic compounds in ambient aerosol particles below the DRH. Most inorganic species are completely dissolved beyond the DRH point in the subsaturated regime, allowing $\kappa_{\text{CCN}}$ to assume complete particle solubility (Pajunoja et al., 2015; Petters & Kreidenweis, 2007).

![Figure 4](image)  
**Figure 4.** Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance (QCM) experiments ($\kappa_m$), and from size-resolved cloud condensation nuclei (CCN) measurements ($\kappa_{\text{CCN}}$). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 – 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by assuming a density of 1.2 g cm$^{-3}$ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived from QCM experiments ($\kappa_m$; orange points) and size-resolved CCN measurements ($\kappa_{\text{CCN}}$; blue points) are shown for the comparison. The error bars for $\kappa_m$ and $\kappa_{\text{CCN}}$ indicate the measurement uncertainty and variability, respectively.

Unlike QCM measurements, the hygroscopicity parameters obtained in previous studies using the HTDMA technique ($\kappa_{\text{HTDMA}}$) are often consistent with $\kappa_{\text{CCN}}$, as the particle
hygroscopicity is determined by particle size increase above DRH point, unaffected by solubility limits (Pajunoja et al., 2015; Y. Wang et al., 2018; Wu et al., 2013). The technical limitations, particularly limited diameter changes corresponding to $g_{f_m}$ values less than 2.0, prevent HTDMA from measuring aerosol hygroscopic properties below the DRH point in the subsaturated regime (Laskina et al., 2015). In contrast, the QCM technique, which is not subject to these limitations, can be used to understand the water uptake characteristics of aerosol particles in the subsaturated regime below DRH and substantially enhance our understanding about solubility limitations of organic and inorganic compounds in the lower RH region. Such an enhancement in the knowledge is crucial for accurately quantifying the radiative forcing effects of ambient aerosol particles in the atmosphere. The strong size dependence of $\kappa_m$ also indicated varying chemical composition for different size ranges measured in this study.

4. SUMMARY AND ATMOSPHERIC IMPLICATIONS

We report the first results based on a high-sensitivity QCM technique to investigate the mass-based growth factor and hygroscopicity parameter of size–resolved ambient aerosols over a wide range of RH from a relatively pristine high–altitude site in India. For the investigated size ranges, the pronounced variations in $g_{f_m}$ below the DRH provided an opportunity to better understand the changes in ambient aerosol properties even at low RH values, which may not be revealed by size-based growth factor measurements (Hu et al., 2010; Laskina et al., 2015). These measurements clearly highlight the important and critical role in knowing the changes in aerosol properties based on their RH history for an improved understanding of water uptake, phase transition, and radiative impact of atmospheric aerosol particles (Zhao et al., 2022). Based on the estimated $\kappa_m$ values, the ambient aerosol particles exhibited the behaviour of an ideal solution (Pajunoja et al., 2015) and strong size-dependent chemical composition. We further observed a pronounced size dependency of the DRH values potentially resulting from complexities of organic solubility in organic–inorganic mixtures (Li et al., 2021). Such a complexity owing to the presence of various organic species may alter the CCN behaviour of atmospheric aerosol particles due to phase transitions and changes in physical properties. The understanding of CCN activation in the supersaturated regime under the assumption of complete solubility of particles is relatively well established. But the understanding of the thermodynamic properties associated with particle growth in the subsaturated regime still remains a challenge. Our findings emphasize the need for additional experiments on ambient and atmospherically relevant laboratory-generated aerosol particles.
using high-sensitivity techniques like QCM. We demonstrated the importance of this technique to better understand the rapid changes in aerosol properties resulting from exposure to a wide range of atmospheric RH conditions. The enhancement in our understanding of the complex interplay between water vapour and aerosol particles will help in developing more accurate models to effectively describe the role of aerosols in atmospheric processes to reduce climate uncertainties and assess the impact of air pollution on human and ecosystem health.
ASSOCIATED CONTENT

Author contributions
SSG conceived the idea. SSG and PL conceptualized the study and designed the research. CJ further developed and validated the QCM measurements for the ambient aerosol studies. CJ performed the field measurement campaign to collect the aerosol samples with support from AS, KNK, and RKA. CJ performed all the laboratory experiments using QCM with support from SS. GVJ performed the satellite data analysis to obtain the AOD values. CJ carried out the data analysis obtained from QCM with input from PL and SSG. CJ and AS performed the scientific interpretation of the QCM data under the mentorship of RR, SSG and PL. CJ wrote the first draft of manuscript under the mentorship of SSG with inputs from PL and RR, and further edits from SY. MOA and STM further provided critical and valuable inputs on the manuscript.

Open research
The data used in the manuscript has been deposited in an open research repository as Excel files, accessible at https://doi.org/10.6084/m9.figshare.24512377. MODIS data were sourced from the Level 2 and Atmosphere Archive and Distribution System (LAADS) (https://doi.org/10.5067/MODIS/MYD04_3K.061). Figures were prepared using Igor Pro version 9 (WaveMetrics Inc.), licensed by SSG and accessible at https://www.wavemetrics.com/software/igor-pro-9.

Acknowledgments
SSG gratefully acknowledges funding from the Ministry of Earth Sciences (MoES; sanction number MoES/16/20/12-RDEAS dated 31. Mar.2014), Government of India, for the purchase of the Cloud Condensation Nuclei Counter (CCNc). This work was supported by partial funding from the Ministry of Earth Sciences (MoES; sanction number MoES/16/04/2017-APHH (PROMOTE)), the Government of India, and the Department of Science and Technology (sanction number DST/CCP/CoE/141/2018C), the Government of India for the purchase of Quartz Crystal Microbalance (QCM). PL acknowledges the start-up funding support from the Georgia Institute of Technology. CJ acknowledges the Department of Science and Technology, the Government of India for the fellowship. Authors acknowledge the valuable support and help provided by the staff at the College of Engineering Munnar during the campaign, with special and critical help from Jyothish Jose. CJ acknowledges the
invaluable assistance provided by Emil Varghese during the laboratory experiments. We are thankful to the support staff from Biolin Scientific, and Specialise Instruments Marketing Company, Mumbai, India for their help during the experiments. We acknowledge the National Aeronautics and Space Administration (NASA) for making Moderate Resolution Imaging Spectroradiometer (MODIS) Earth data available to the user community.
References


Liu, Y., & Laskin, A. (2009). Hygroscopic Properties of CH₃SO₃Na, CH₃SO₃NH₄, (CH₃SO₃)₂Mg, and (CH₃SO₃)₂Ca Particles Studied by micro-FTIR Spectroscopy. 1531–1538.


Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon season of 2021 (June - September). The AOD distribution clearly indicates a relatively lower aerosol loading over the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind rose diagram shown in the inset is average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August – September 2021). The prevailing air masses mostly originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the observational site, confirming the relatively low influence of anthropogenic activities.
Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For panels a, b, and c, $\Delta f_N$ represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) conditions normalised to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in the value of $\Delta f_N$ for each size range for the sampled ambient aerosol particles indicates the water uptake at different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size ranges. In panels d, e, and f the derivative of $\Delta f_N$ with respect to RH ($d(\Delta f_N)/d(RH)$) is plotted against RH to determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. The RH values at which $d(\Delta f_N)/d(RH)$ becomes $\geq 1$ (marked by the dotted line) represent the DRH values for the individual aerosol size ranges.
Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor ($g_{fm}$) derived using a quartz crystal microbalance (QCM) for ten different size ranges of ambient aerosol particles at different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the corresponding $\kappa$-Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity parameter, $\kappa_m$. The error bars represent the variations in $g_{fm}$ averaged over the mass change corresponding to different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the mean $\kappa_m$ values corresponding to respective size ranges. (b) The data points are the $\kappa_m$ values calculated based on the $g_{fm}$ (as shown in (a)) using the $\kappa$-Köhler theory for different RH conditions in the subsaturated regime. The dotted lines represent the mean $\kappa_m$ value for each size range of ambient aerosol particles and the error bars represent one standard deviation.
Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance (QCM) experiments ($\kappa_m$), and from size-resolved cloud condensation nuclei (CCN) measurements ($\kappa_{CCN}$). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 – 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by assuming a density of 1.2 g cm$^{-3}$ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived from QCM experiments ($\kappa_m$; orange points) and size-resolved CCN measurements ($\kappa_{CCN}$; blue points) are shown for the comparison. The error bars for $\kappa_m$ and $\kappa_{CCN}$ indicate the measurement uncertainty and variability, respectively.
Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique


1Environmental Engineering Division, Dept of Civil Engineering, Indian Institute of Technology Madras, Chennai 600036, India
2Centre for Atmospheric and Climate Sciences, Indian Institute of Technology Madras, Chennai 600036, India
3Dept of Civil Engineering, Indian Institute of Technology Bombay, Mumbai 400076, India
4Dept of Mechanical Engineering, College of Engineering Munnar, Munnar 685612, India
5Dept of Environmental Sciences, Central University of Jammu, Samba, Jammu and Kashmir 181143, India
6Dept of Chemical Engineering, Indian Institute of Technology Madras, Chennai 600036, India
7Multiphase Chemistry Department, Max Planck Institute for Chemistry, 55128 Mainz, Germany
8Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA 92093, USA
9Department of Geology and Geophysics, King Saud University, Riyadh 11451, Saudi Arabia
10Department of Earth and Planetary Sciences, Harvard University, Cambridge, MA 02138, USA
11John A. Paulson School of Engineering & Applied Sciences, Harvard University, Cambridge, MA 02138, USA
12School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA 30332, USA

*Correspondence to: Sachin S. Gunthe (s.gunthe@iitm.ac.in) and Pengfei Liu (pengfei.liu@eas.gatech.edu)

KEYPOINTS

- QCM technique revealed aerosol properties at low RH values offering insights that may not be captured by traditional size-based measurements
- The size-dependent variations in aerosol properties below DRH emphasize the need to understand how aerosol properties change with RH history
- Organic in aerosols complicate its cloud forming ability, necessitating additional studies in diverse environments to improve climate models
ABSTRACT
Comprehending the intricate interplay between atmospheric aerosols and water vapour in subsaturated regions is vital for accurate modelling of aerosol–cloud–radiation–climate dynamics. But the microphysical mechanisms governing these interactions with ambient aerosols remain inadequately understood. Here we report results from high-altitude, relatively pristine site in Western-Ghats of India during monsoon, serving as a baseline for climate processes in one of the world’s most polluted regions. Utilizing a novel quartz crystal microbalance (QCM) approach, we conducted size-resolved sampling to analyse humidity-dependent growth factors, hygroscopicity, deliquescence behaviour, and aerosol liquid water content (ALWC). Fine-mode aerosols (≤2.5 μm) exhibited size-dependent interactions with water vapour, contributing significantly to ALWC. Deliquescence was observed in larger aerosols (>180 nm), influenced by organic species, with deliquescence relative humidity (DRH) lower than that of pure inorganic salts. This research highlights the significance of understanding ambient aerosol-water interactions and hygroscopicity for refining climate models in subsaturated conditions.

PLAIN LANGUAGE SUMMARY
Aerosol particles interact with water vapour in the atmosphere. Understanding these interactions in sub – and super-saturated regions is crucial because they affect processes such as cloud formation, radiation, and climate. We collected ambient aerosol samples from Western Ghats in India to understand natural processes that happen without significant human interventions. Using a special technique involving a quartz crystal sensor, we measured different aspects, such as how aerosols grow with humidity, their ability to uptake the water molecules, and when they transition from solid to liquid (deliquescence). Interestingly, fine particles, especially those smaller than 2.5 μm, exhibited unique behaviours. While larger particles underwent a transition from solid to liquid under certain conditions, this didn’t happen for the smaller particles. These findings highlight the importance of understanding these interactions for more accurate climate predictions.
1. INTRODUCTION

Understanding the hygroscopicity of atmospheric aerosols is crucial for assessing cloud formation and their climate and air pollution impact (Cheung et al., 2015). Investigation of the hygroscopicity of ambient aerosols poses significant challenges due to their chemical complexity, particle size variation, phase state, and viscosity. Techniques like Fourier transform infrared spectroscopy (FTIR) (Y. Liu et al., 2008; Y. Liu & Laskin, 2009), quartz crystal microbalance (QCM) (Chao et al., 2020; Demou et al., 2003; P. Liu et al., 2016a; P. Liu, Song, et al., 2018a), Raman spectroscopy (Ling & Chan, 2008; Y. J. Liu et al., 2008), electrodynamic balance (EDB) (Choi & Chan, 2002; Peng & Chan, 2001; Pope et al., 2010), optical microscopy (OM) (Ahn et al., 2010; Eom et al., 2014; Gupta et al., 2015), hygroscopicity tandem differential mobility analysis (HTDMA) (Cheung et al., 2015; Prenni et al., 2007; Zieger et al., 2017a) and size-selected cloud condensation nuclei (CCN) spectrometry (Petters et al., 2007; Pöhlker et al., 2016; Rose et al., 2008) have been employed to study the hygroscopicity of laboratory – generated and ambient aerosols (Tang et al., 2019). However, a consensus on the most effective method is lacking, leading to inconsistent and incomparable results. Hygroscopicity measurements primarily focus on accumulation mode particles, limiting data on nucleation and coarse mode particles with complex behaviours. Discrepancies in hygroscopic properties and organic and inorganic species composition in atmospheric aerosols result in diverse growth factors and phase transitions, crucial for accurate climate modelling (Li et al., 2021).

The QCM is effective in determining mass–based hygroscopicity and physical property variations of atmospherically–relevant aerosols (Demou et al., 2003; P. Liu et al., 2016b; P. Liu, Song, et al., 2018b). It overcomes the limitations of traditional methods like HTDMA, which has a narrow particle size range with limited relative humidity (RH) resolution (Zhao et al., 2022). Conventional HTDMA methods inadequately reveal hygroscopicity heterogeneity and phase transitions in wider aerosol size range, impacting processes like CCN activation and aerosol liquid water (Li et al., 2021). QCM’s mass-based measurements accurately quantify aerosol hygroscopicity behaviour across a wide RH range in the atmosphere, detailing physical property variations during phase transitions and help to delineate water uptake mechanisms, including solubility and water diffusion limitations (P. Liu, Li, et al., 2018; P. Liu, Song, et al., 2018b).
India's climate is distinct and intricate, and as per the 2020 climate change assessment report, India has faced challenges like temperature rise and extreme weather events since the mid-20th century. A key hurdle in addressing these issues is the absence of systematic aerosol characteristic measurements, particularly in aerosol-water vapor interactions in the subsaturated regime (Cheung et al., 2015). Aerosol particles from a forest-surrounded by pristine area are anticipated to contain substantial biogenic organic aerosols. The dynamic chemical transformation of secondary organic aerosols (SOA) in the atmosphere leads to intricate compositions. Atmospheric processing further complicates the understanding of phase state and properties of atmospheric aerosols (Rastak et al., 2017). In this study, QCM measurements are utilized to explore aerosol hygroscopicity dependence on RH, chemical composition, and size across a broad range of ambient aerosols in India’s relatively pristine setting.

2. MATERIALS AND METHODS

Size-resolved ambient aerosols were sampled at the Natural Aerosol and Bioaerosol High Altitude (NABHA) Laboratory at the College of Engineering, Munnar (10.0930° N, 77.0682° E; 1600 m above mean sea level), a high-altitude and typically clean site situated in the Western Ghats of India, using a 10-stage micro-orifice uniform deposit impactor (MOUDI-II 120-R, TSI) (V. Marple et al., 2014; V. A. Marple et al., 1991) during the monsoon season (August-September) of 2021. Ambient particles collected on PTFE (Polytetrafluoroethylene) filters were transferred to a hydrophobic SiO₂-coated quartz sensor by gently pressing the filter paper onto the sensor. The hygroscopic growth factor, hygroscopicity parameter and deliquescence relative humidity (DRH) were estimated using a highly sensitive mass balance instrument, QCM (QSense Analyzer, Biolin Scientific) (P. Liu, Song, et al., 2018a; Reviakine et al., 2011). The accuracy and robustness of the method were demonstrated by performing measurements of the hygroscopic growth factor and DRH for sucrose and (NH₄)₂SO₄ particles, respectively, which were compared with previously reported results (Arenas et al., 2012a; Chao et al., 2020; Martin, 2000; Norrish, 1966; Peng et al., 2022; Starzak & Peacock, 1997; Zobrist et al., 2011) (Fig. S3 and S4). The hygroscopic growth factor and the corresponding hygroscopicity parameter, κ, were determined over a wide range of RH conditions using κ-Köhler theory (Petters & Kreidenweis, 2007), where κ represents a quantitative measure of aerosol water uptake characteristics and CCN activity (Zhao et al., 2022). The water uptake characteristics and the phase transition behaviours of the ambient
samples were investigated and compared between different size ranges of particles below 10 μm collected using the MOUDI sampler. The Aerosol Liquid Water Content (ALWC) (Bian et al., 2014) at different RH conditions was also estimated from corresponding hygroscopicity parameters for each size range of particles (Fig. S5). The number size distributions of the ambient aerosol particles of size 10 to 420 nm [measured in parallel using a scanning mobility particle sizer – SMPS, comprised of an electrostatic classifier (EC, TSI Model 3082) equipped with a differential mobility analyser (DMA, TSI Model 3081) and a condensation particle counter (CPC, TSI Model 3750)], were then converted to mass size distributions assuming a particle density of 1.2 g cm\(^{-3}\) (DeCarlo et al., 2004). The supermicron particle mass size distribution for the same season, but a different year, was used in this study, which had been obtained using an Ultraviolet Aerodynamic Particle Sizer (UV-APS, TSI Inc., model 3314) (Valsan et al., 2016).

Meteorological parameters were recorded using an automatic weather station (AWS, Clima Sensor US) during the sampling period. Figure 1 shows the sampling site – Munnar, along with aerosol optical depth (AOD) over India during the monsoon season, indicating the relatively pristine nature of the sampling site. During the measurement, airmasses predominantly arrived from the southwest direction, bringing clean marine influx to the observational site and resulting in a low influence of anthropogenic emissions (Fig. 1). A more detailed description of the instruments, experimental techniques, and estimation of various parameters is provided in the Supporting Information (SI).
Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon season of 2021 (June - September). The AOD distribution clearly indicates a relatively low aerosol loading over the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind rose diagram shown in the inset shows the average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August – September 2021). The prevailing air masses mostly originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the observational site, confirming the relatively low influence of anthropogenic activities.

3. RESULTS AND DISCUSSION

The QCM sensor’s oscillation frequency variation (Δf) signifies water uptake and release, providing insights into adsorption, desorption, and physical states of particles during solid-to-aqueous phase transition (Arenas et al., 2012b). Normalizing the frequency shift at higher RH to that of deposited dry sample (RH <5%) yields the percentage value, Δf_N (Chao et al., 2020). Figure 2 shows Δf_N for the ambient aerosol particles across various RH levels and sampled size ranges. A negative sign in Δf_N indicates frequency reduction with increasing RH due to water uptake. Further details of Δf_N and its derivative, d(Δf_N)/(RH), are available in SI.

For the particle size <180 nm (Fig. 2a), Δf_N decreased, implying water uptake, with <56 nm particles exhibiting the lowest value. In Figure 2b and 2c, particles showed increased water uptake at lower RH, evidenced by Δf_N decrease. However, Δf_N suddenly increased at specific RH, indicating a drastic water uptake leading to deliquescence-induced phase transition. Figures 2a-c revealed deliquescence in particles >180 nm at different RH (known as DRH), absent in those <180 nm. Past studies reported no deliquescence for ambient and laboratory-generated atmospherically relevant particles even at high RH, attributing them to organic nature (Arenas et al., 2012a; Brooks et al., 2002; Chao et al., 2020; Peng et al., 2022). We hypothesize that particles <180 nm were likely dominated by freshly formed secondary organic aerosols (SOA) from biogenic volatile organic compound (VOC) oxidation, supported by the densely vegetated/forest region and season. Concurrent quasi-continuous measurements during the same campaign indicated upto ~90% organic fraction in NR-PM₁ (non-refractory particulate matter with an aerodynamic diameter ≤1 μm; Tab. S2) with details discussed in subsequent studies.
Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For panels a, b, and c, $\Delta f_N$ represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) conditions normalized to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in the value of $\Delta f_N$ for each size range for the sampled ambient aerosol particles indicates the water uptake at different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size ranges. In panels d, e, and f, the derivative of $\Delta f_N$ with respect to RH ($d(\Delta f_N)/d(RH)$) is plotted against RH to determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. The RH values at which $d(\Delta f_N)/d(RH)$ becomes ≥1 (marked by the dotted line) represent the DRH values for the individual aerosol size ranges.

Following Chao et al. (2020), DRH, indicated by $d(\Delta f_N)/(RH) \geq 1$ ranged from ~60 – 68% for three size ranges (180 nm – 320 nm; 320 nm – 560 nm; 560 nm – 1 µm) in this study (Fig. 2c). Above 180 nm, inorganic salts, like (NH$_4$)$_2$SO$_4$, may contribute to DRH appearance (Hu et al., 2010) (Fig. 2b,e). These DRH values, smaller than pure (NH$_4$)$_2$SO$_4$ (DRH=80%), suggest mixed salts and/or organic compounds in the sampled aerosol particles. The deliquescence behaviour of organic-inorganic mixtures introduces complexity due to solubility limitations, influenced by the organic species in the aerosol particles. Previous studies noted reduced DRH of inorganic species and sometimes, unaffected by organics (Smith et al., 2011, 2012, 2013). Thus, we hypothesize 180 nm to 1 µm particles at this site reflect a complex mixture of inorganic salts like (NH$_4$)$_2$SO$_4$ and water-soluble organic compounds. Supermicron particles (>1 µm), except 1 – 1.8 µm, showed <17% $\Delta f_N$, indicating lower water uptake (Fig. 2c). Beyond maximum water uptake, the three size ranges exhibited different DRH values. 1.8 – 3.2 µm and 5.6 – 10 µm exhibited DRH between 56 – 68%, suggesting a mixture of water-soluble organic compounds and inorganic salts. However, 3.2 – 5.6 µm showed a very high DRH (>81%) with the organic fraction minimally affecting DRH due to solubility limitations. For 1.8 – 3.2 µm and 5.6 – 10 µm, reduced DRH
compared to pure inorganic salts suggests a mix of water-soluble organic and inorganic compounds (Smith et al., 2011, 2012, 2013). The 1.0 – 1.8 μm range exhibited a >30% ΔfN reduction at ~42% RH, indicating a higher water uptake and the observed DRH appeared to be ~45 – 53%, consistent with the highest κm (~0.44, see discussion below) among all measured size ranges (Chao et al., 2020). The low DRH at this site suggests presence of highly water-soluble organic material within the organic-inorganic (sea salt) mixture in this size range.

Figure 3 shows the size-resolved mass-based hygroscopic growth factors (gf_m) and the corresponding mass-based hygroscopicity parameters (κm) of ambient aerosol particles calculated from QCM measurements (Text S4 and S5) over a wide RH range (2 – 90%). In Figure 3a, the growth factor increases with water uptake in humid air, showing size-dependent variations across different RH levels (indicated by markers). Each size range had a maximum RH limit beyond which gf_m estimation was impossible due to deliquescence, as observed in previous studies (Chao et al., 2020). Accordingly, the highest gf_m=1.72 was recorded at RH=90% for the 100 – 180 nm particles. The pronounced variations in gf_m between different-sized particles at a given RH indicate varying chemical composition and hygroscopicity across the sampled sizes.

Figure 3b shows the size-dependent average κm values, calculated from the gf_m values in Figure 3a. Each particle size range exhibited minimal κm variation (average κm ± 0.007), suggesting ideal solution behaviour across the wide RH range (Z. Wang et al., 2017). The average κm values (indicated by dashed lines in Figure 3b) of the size-resolved ambient particles varied significantly between 0.016 (<0.056 nm) and 0.44 (1.0 – 1.8 μm) over the investigated RH range. This implies κm as a function of size, and indicates distinct chemical compositions for each size fraction. For example, the small increase in κm from 0.016 to 0.078 for particles <56 nm to 56-100 nm, respectively, is potentially owing to aging processes (Zhang et al., 2023). Aerosol particles <320 nm exhibited very low average κm (0.06), which may be primarily due to the strong dominance of organic compounds in these size ranges (Demou et al., 2003; Li et al., 2021; Shi et al., 2022). The particles in the size range of 320 nm – 1 μm were moderately hygroscopic (average κm=0.19) indicating the influence of inorganic salts (Wu et al., 2016). The particles between 1.0 – 1.8 μm exhibited highest κm (0.44), which implies the presence of highly hygroscopic material such as sea salt.
(Zieger et al., 2017b) mixed with organic compounds. Even larger particles exhibited reduced \( \kappa_m \), possibly due to the presence of dust particles (Koehler et al., 2009), transported at this site during monsoon season (Valsan et al., 2016). The \( \kappa \)-Köhler parameterization fit accurately elucidates the \( g_{fm} \) within each size range over a wide range of RH values (Petters & Kreidenweis, 2007).

Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor (\( g_{fm} \)) derived using a quartz crystal microbalance (QCM) for ten different size ranges of ambient aerosol particles at different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the corresponding \( \kappa \)-Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity parameters, \( \kappa_m \). The error bars represent the variations in \( g_{fm} \) averaged over the mass change corresponding to different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the mean \( \kappa_m \) values corresponding to the respective size ranges. (b) The data points are the \( \kappa_m \) values calculated based on the \( g_{fm} \) (as shown in (a)) using the \( \kappa \)-Köhler theory for different RH conditions in the subsaturated regime. The dotted lines represent the mean \( \kappa_m \) value for each size range of ambient aerosol particles and the error bars represent one standard deviation.

Figure 4 shows the size dependency of \( \kappa_m \) in the subsaturated regime measured by QCM together with \( \kappa_{CCN} \) (Text S2) in the supersaturated regime determined by size-resolved CCN measurements and aerosol size distributions obtained by SMPS and UV-APS. The submicron particles show a bimodal number size distribution with an Aitken mode peak at 63 nm and an accumulation mode at 145 nm, while the mass size distributions peak at 350 nm and 3 \( \mu \)m.
respectively. The average $\kappa_{\text{CCN}}$ increased from a value of 0.28 for the Aitken mode particles to 0.47 for the accumulation mode particles. In the submicron region, $\kappa_m$ exhibited a nominal increase with particle size (Wu et al., 2016), in parallel with the increase in $\kappa_{\text{CCN}}$, albeit at much lower absolute values. This increase likely reflects the decrease of organic fraction with size in the submicron range, which is supported by the appearance of deliquescence in the size range above 180 nm, indicating the presence of inorganic salts. In the supermicron region (for particles $\geq 1 \mu m$), $\kappa_m$ peaked due to the potential presence of NaCl particles in the size range 1.0 – 1.8 $\mu m$, and decreased again in the higher size ranges (>1.8 $\mu m$) likely because of the presence of dust particles. The average $\kappa_m$ value (0.18) obtained from QCM measurements across all the sampled size ranges in the subsaturated regime was lower than the average $\kappa_{\text{CCN}}$ (0.39), likely due to the solubility limitation (Dusek et al., 2011; Hersey et al., 2013; Rastak et al., 2017; Riipinen et al., 2015; Wittbom et al., 2018) of organic and inorganic compounds in ambient aerosol particles below the DRH. Most inorganic species are completely dissolved beyond the DRH point in the subsaturated regime, allowing $\kappa_{\text{CCN}}$ to assume complete particle solubility (Pajunoja et al., 2015; Petters & Kreidenweis, 2007).

Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance (QCM) experiments ($\kappa_m$), and from size-resolved cloud condensation nuclei (CCN) measurements ($\kappa_{\text{CCN}}$). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 – 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by assuming a density of 1.2 g cm$^{-3}$ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived from QCM experiments ($\kappa_m$; orange points) and size-resolved CCN measurements ($\kappa_{\text{CCN}}$; blue points) are shown for the comparison. The error bars for $\kappa_m$ and $\kappa_{\text{CCN}}$ indicate the measurement uncertainty and variability, respectively.

Unlike QCM measurements, the hygroscopicity parameters obtained in previous studies using the HTDMA technique ($\kappa_{\text{HTDMA}}$) are often consistent with $\kappa_{\text{CCN}}$, as the particle
hygroscopicity is determined by particle size increase above DRH point, unaffected by solubility limits (Pajunoja et al., 2015; Y. Wang et al., 2018; Wu et al., 2013). The technical limitations, particularly limited diameter changes corresponding to $g_{fm}$ values less than 2.0, prevent HTDMA from measuring aerosol hygroscopic properties below the DRH point in the subsaturated regime (Laskina et al., 2015). In contrast, the QCM technique, which is not subject to these limitations, can be used to understand the water uptake characteristics of aerosol particles in the subsaturated regime below DRH and substantially enhance our understanding about solubility limitations of organic and inorganic compounds in the lower RH region. Such an enhancement in the knowledge is crucial for accurately quantifying the radiative forcing effects of ambient aerosol particles in the atmosphere. The strong size dependence of $\kappa_m$ also indicated varying chemical composition for different size ranges measured in this study.

4. SUMMARY AND ATMOSPHERIC IMPLICATIONS

We report the first results based on a high-sensitivity QCM technique to investigate the mass-based growth factor and hygroscopicity parameter of size–resolved ambient aerosols over a wide range of RH from a relatively pristine high–altitude site in India. For the investigated size ranges, the pronounced variations in $g_{fm}$ below the DRH provided an opportunity to better understand the changes in ambient aerosol properties even at low RH values, which may not be revealed by size-based growth factor measurements (Hu et al., 2010; Laskina et al., 2015). These measurements clearly highlight the important and critical role in knowing the changes in aerosol properties based on their RH history for an improved understanding of water uptake, phase transition, and radiative impact of atmospheric aerosol particles (Zhao et al., 2022). Based on the estimated $\kappa_m$ values, the ambient aerosol particles exhibited the behaviour of an ideal solution (Pajunoja et al., 2015) and strong size-dependent chemical composition. We further observed a pronounced size dependency of the DRH values potentially resulting from complexities of organic solubility in organic–inorganic mixtures (Li et al., 2021). Such a complexity owing to the presence of various organic species may alter the CCN behaviour of atmospheric aerosol particles due to phase transitions and changes in physical properties. The understanding of CCN activation in the supersaturated regime under the assumption of complete solubility of particles is relatively well established. But the understanding of the thermodynamic properties associated with particle growth in the subsaturated regime still remains a challenge. Our findings emphasize the need for additional experiments on ambient and atmospherically relevant laboratory-generated aerosol particles.
using high-sensitivity techniques like QCM. We demonstrated the importance of this technique to better understand the rapid changes in aerosol properties resulting from exposure to a wide range of atmospheric RH conditions. The enhancement in our understanding of the complex interplay between water vapour and aerosol particles will help in developing more accurate models to effectively describe the role of aerosols in atmospheric processes to reduce climate uncertainties and assess the impact of air pollution on human and ecosystem health.
ASSOCIATED CONTENT

Author contributions

SSG conceived the idea. SSG and PL conceptualized the study and designed the research. CJ further developed and validated the QCM measurements for the ambient aerosol studies. CJ performed the field measurement campaign to collect the aerosol samples with support from AS, KNK, and RKA. CJ performed all the laboratory experiments using QCM with support from SS. GVJ performed the satellite data analysis to obtain the AOD values. CJ carried out the data analysis obtained from QCM with input from PL and SSG. CJ and AS performed the scientific interpretation of the QCM data under the mentorship of RR, SSG and PL. CJ wrote the first draft of manuscript under the mentorship of SSG with inputs from PL and RR, and further edits from SY. MOA and STM further provided critical and valuable inputs on the manuscript.

Open research

The data used in the manuscript has been deposited in an open research repository as Excel files, accessible at https://doi.org/10.6084/m9.figshare.24512377. MODIS data were sourced from the Level 2 and Atmosphere Archive and Distribution System (LAADS) (https://doi.org/10.5067/MODIS/MYD04_3K.061). Figures were prepared using Igor Pro version 9 (WaveMetrics Inc.), licensed by SSG and accessible at https://www.wavemetrics.com/software/igor-pro-9.

Acknowledgments

SSG gratefully acknowledges funding from the Ministry of Earth Sciences (MoES; sanction number MoES/16/20/12-RDEAS dated 31. Mar.2014), Government of India, for the purchase of the Cloud Condensation Nuclei Counter (CCNc). This work was supported by partial funding from the Ministry of Earth Sciences (MoES; sanction number MoES/16/04/2017-APHH (PROMOTE)), the Government of India, and the Department of Science and Technology (sanction number DST/CCP/CoE/141/2018C), the Government of India for the purchase of Quartz Crystal Microbalance (QCM). PL acknowledges the start-up funding support from the Georgia Institute of Technology. CJ acknowledges the Department of Science and Technology, the Government of India for the fellowship. Authors acknowledge the valuable support and help provided by the staff at the College of Engineering Munnar during the campaign, with special and critical help from Jyothish Jose. CJ acknowledges the
invaluable assistance provided by Emil Varghese during the laboratory experiments. We are thankful to the support staff from Biolin Scientific, and Specialise Instruments Marketing Company, Mumbai, India for their help during the experiments. We acknowledge the National Aeronautics and Space Administration (NASA) for making Moderate Resolution Imaging Spectroradiometer (MODIS) Earth data available to the user community.
References


Liu, Y., & Laskin, A. (2009). Hygroscopic Properties of CH₃SO₃Na, CH₃SO₃NH₄, (CH₃SO₃)₂Mg, and (CH₃SO₃)₂Ca Particles Studied by micro-FTIR Spectroscopy. 1531–1538.


Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon season of 2021 (June - September). The AOD distribution clearly indicates a relatively lower aerosol loading over the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind rose diagram shown in the inset is average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August – September 2021). The prevailing air masses mostly originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the observational site, confirming the relatively low influence of anthropogenic activities.
Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For panels a, b, and c, $\Delta f_N$ represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) conditions normalised to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in the value of $\Delta f_N$ for each size range for the sampled ambient aerosol particles indicates the water uptake at different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size ranges. In panels d, e, and f the derivative of $\Delta f_N$ with respect to RH ($d(\Delta f_N)/d(RH)$) is plotted against RH to determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. The RH values at which $d(\Delta f_N)/d(RH)$ becomes $\geq 1$ (marked by the dotted line) represent the DRH values for the individual aerosol size ranges.
Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor ($g_f$) derived using a quartz crystal microbalance (QCM) for ten different size ranges of ambient aerosol particles at different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the corresponding $\kappa$-Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity parameter, $\kappa_m$. The error bars represent the variations in $g_f$ averaged over the mass change corresponding to different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the mean $\kappa_m$ values corresponding to respective size ranges. (b) The data points are the $\kappa_m$ values calculated based on the $g_f$ (as shown in (a)) using the $\kappa$-Köhler theory for different RH conditions in the subsaturated regime. The dotted lines represent the mean $\kappa_m$ value for each size range of ambient aerosol particles and the error bars represent one standard deviation.
Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance (QCM) experiments ($\kappa_m$), and from size-resolved cloud condensation nuclei (CCN) measurements ($\kappa_{CCN}$). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 – 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by assuming a density of 1.2 g cm$^{-3}$ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived from QCM experiments ($\kappa_m$; orange points) and size-resolved CCN measurements ($\kappa_{CCN}$; blue points) are shown for the comparison. The error bars for $\kappa_m$ and $\kappa_{CCN}$ indicate the measurement uncertainty and variability, respectively.
Supporting Information

Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique

Christi Jose¹², Aishwarya Singh¹², Kavyashree N. Kalkura¹², George V. Jose³, Shailina Srivastava¹², Rameshchand K. A. ⁴, Shweta Yadav⁵, R. Ravikrishna²⁶, M. O. Andreae⁷⁸⁹, Scot T. Martin¹⁰¹¹, Pengfei Liu¹²*, and Sachin S. Gunthe¹²*

¹Environemntal Engineering Division, Dept of Civil Engineering, Indian Institute of Technology Madras, Chennai 600036, India
²Centre for Atmospheric and Climate Sciences, Indian Institute of Technology Madras, Chennai 600036, India
³Dept of Civil Engineering, Indian Institute of Technology Bombay, Mumbai 400076, India
⁴Dept of Mechanical Engineering, College of Engineering Munnar, Munnar 685612, India
⁵Dept of Environmental Sciences, Central University of Jammu, Samba, Jammu and Kashmir 181143, India
⁶Dept of Chemical Engineering, Indian Institute of Technology Madras, Chennai 600036, India
⁷Multiphase Chemistry Department, Max Planck Institute for Chemistry, 55128 Mainz, Germany
⁸Scripps Institution of Oceanography, University of California San Diego, La Jolla, CA 92093, USA
⁹Department of Geology and Geophysics, King Saud University, Riyadh 11451, Saudi Arabia
¹⁰Department of Earth and Planetary Sciences, Harvard University, Cambridge, MA 02138, USA
¹¹John A. Paulson School of Engineering & Applied Sciences, Harvard University, Cambridge, MA 02138, USA
¹²School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA 30332, USA

*Correspondence to: Sachin S. Gunthe (s.gunthe@iitm.ac.in) and Pengfei Liu (pengfei.liu@eas.gatech.edu)
Text S1. Sampling Site

The sampling was conducted at the Natural Aerosol and Bioaerosol High Altitude (NABHA) Laboratory at the College of Engineering, Munnar (10.0930° N, 77.0682° E), located in the Western Ghats of India, one of the most significant biodiversity hotspots in the world (Kale et al., 2016). The site is at about ~1600 m in elevation and 90 km from the Arabian Sea. Munnar possesses a moderately rugged topography with high mountain peaks and deep river valleys covered with diversified vegetated areas, tea plantations, barren land/rocky areas, and water bodies (Gupta et al., 2020). During the southwest monsoon season (June-September), the Western Ghats block the south-westerly winds, causing rainfall on these mountain ranges.

Text S2. Instrumentation

The ambient aerosols were collected on polytetrafluoroethylene (PTFE) membrane filters (pore size 0.2 μm) using a Micro-Orifice Uniform Deposit Impactor (MOUDI) with a sampling flow rate of 30 L/min. The MOUDI-II 120R (TSI) utilized 10 rotating stages to achieve a uniform deposit of particles on the filters, with nominal cut-off points at 0.056, 0.10, 0.18, 0.32, 0.56, 1.0, 1.8, 3.2, 5.6, and 10 μm (Marple et al., 2014). Concurrently, various meteorological parameters such as temperature, humidity, pressure, wind speed, wind direction, and precipitation were recorded using an Automatic Weather Station (AWS, Clima Sensor US) at the sampling location throughout the sampling period.

The number size distribution of the ambient aerosol particles was measured using a Scanning Mobility Particle Sizer (SMPS), which comprises a 3082 Electrostatic Classifier with 3081 Long Differential Mobility Analyzer (L-DMA, TSI Model 3081) and a butanol-Condensation Particle Counter (CPC, TSI Model 3750). The measurements were carried out with an aerosol flow rate of 1 L/min, spanning a size range of 10.2 – 430 nm. The aerosol mass distribution measurements for the larger-sized particles (0.3 – 20.0 μm) obtained using an Ultraviolet Aerodynamic Particle Sizer (UV-APS, TSI Inc., Model 3314) from a campaign conducted at
the same location during the Monsoon season of 2014 (June-August) were also incorporated in the study.

The hygroscopicity of aerosol particles was analyzed using a sensitive mass-balance Quartz Crystal Microbalance (QCM, QSense Analyzer, Biolin Scientific), which can measure the mass changes at the nanograms level. In 1959, Sauerbrey proposed that an adsorbed mass ($\Delta m$) causes a decrease in the resonant frequency ($\Delta f$) of the quartz crystal, and their linear relationship is given by,

$$\Delta f = -\frac{c}{n} \Delta m$$  \hspace{1cm} (S1)

where $C (=17.7 \text{ ng.cm}^{-2}.\text{Hz}^{-1}$ for 5 MHz AT-cut crystals) is the mass sensitivity constant specific to the quartz resonator and $n (=1,3,5,…)$ is the overtone number (Sauerbrey, 1959). The Sauerbrey equation is valid only when the film deposited on the QCM quartz sensor is rigid and perfectly coupled to the sensor surface (Rodahl & Kasemo, 1995). QCM experiments were performed with a 5 MHz AT-cut quartz sensor coated with SiO$_2$, and the particles were uniformly deposited on the sensor. The ambient aerosol particles collected on PTFE filters were directly transferred to the QCM sensor by placing the filter on the sensor and then gently pressing it using a cotton piece. The particles on the sensor were examined under the microscope to ensure the uniformity of the film.

The relative humidity (RH) was incrementally increased from 2% to 93% to promote the uptake of water by aerosol particles, resulting in a corresponding increase in mass. The QCM provides the fundamental frequencies corresponding to the mass deposited on the sensor. Measurements were taken at various RH points in the sub-saturated region for a blank sensor and a sensor coated with aerosol particles. The difference in the frequencies between the coated and the blank sensor at a particular RH was subsequently converted to the mass of water absorbed or adsorbed at that point using the Sauerbrey equation.

The effective particle hygroscopicity, $\kappa_{\text{CCN}}$ (Petters & Kreidenweis, 2007) under supersaturated conditions was determined using size-resolved Cloud Condensation Nuclei (CCN) measurements acquired through a CCN counter (CCNC, CCN-100, DMT), employing the method outlined in Rose et al. (2008).

**Text S3. Experimental setup**

3
The schematic experimental setup designed to measure the hygroscopicity of the aerosol particles using QCM is depicted in Figure S1. Dry nitrogen was passed through a HEPA filter (Whatman 6702-9500 HEPA-CAP filters) and divided into two channels, one of which was humidified using a water bubbler. Desired RH values were created by controlling the flow ratio of dry and humidified air from both channels using digital mass flow controllers (MFCs) (Alicat Scientific–50 SCCM). The RH was continuously monitored at various points in the flow pathway using thermohydrometers (testo 605i). The accuracy of the RH values in the ranges 10-35%, 35-65%, 65-90% and <10% or >90% are ±3.0%, ±2.0%, ±3.0% and ±5%, respectively. Each humidity condition was sustained until an equilibrium condition was reached, where no further evaporation and condensation happened. The amount of water absorbed by the particles at different values of RH in the sub-saturated region was measured continuously by the QCM. The QCM module temperature was kept at least 0.5 °C higher than the room temperature to avoid condensation of water vapour onto the particles inside the QCM module. The experimental setup employed a low flow range of 0-50 sccm, and any variations in the flow could significantly affect the experimental results. To mitigate this issue, the flow rate was checked at various control points before and after each experiment using a Gilibrator (Sensidyne).

To ensure accurate measurements, it is crucial to properly clean the sensor surface before depositing the particles. The sensors were cleaned with milli-Q water and methanol, then dried with N₂ gas. Subsequently, a UV-Ozone treatment was applied to eliminate various contaminants, particularly organic compounds, from the SiO₂-coated sensor. The treatment was done for 20 minutes at a power of 1 mW/cm² immediately before depositing the sample particles on the sensor.

Text S4. Mass-based hygroscopic growth factor (gfₚ) for laboratory-generated particles

The QCM technique for determining hygroscopicity was validated for a sucrose thin film. The mass-based hygroscopic growth factor (gfₚ, mass ratio of the film at an elevated RH to that at <5% RH) for sucrose particles for a range of RH (2-93%) was calculated from QCM measurements, and the results were compared with the models for testing the accuracy of the method (Liu et al., 2018).

The steps followed for making a sucrose thin film on the sensor are as follows.

1. Prepared different concentrations (0.05%, 0.15, 0.25%, 0.5% etc) of sucrose solution by dissolving sucrose in milli-Q water.
2. Cleaned the sensor surface using Milli-Q water and methanol and treated with UV-Ozone for 20 minutes.

3. The desired volume of the solution was pipetted onto the sensor surface, which was placed on a clean petriplate.

4. The sensor was then placed inside a muffle furnace with a preset temperature of 80°C for 20 minutes to remove water.

5. The sensor with the film was viewed under a microscope (Dino-Lite Edge) to analyze the uniformity of the film.

6. The sensor with a thin film of sucrose was then mounted inside one of the modules of the QCM analyzer.

7. Dry air was allowed to pass over the film for 10-12 hours to further remove the moisture content in the particles.

After checking the uniformity of the film and the stability of different overtones using QSoft software (Fig. S2), the experiment was started with a preset program in FlowVision (software that helps to control the flow rates at the MFCs). The flow rates were changed accordingly in the two MFCs for creating different humidity values starting from dry conditions (<5% RH) to a maximum of 93% RH.

The hygroscopic growth factor values obtained from the QCM measurements were in good agreement with the literature (Norrish, 1966; Starzak & Peacock, 1997; Zobrist et al., 2011) (Fig. S3).

**Text S5. Mass-based hygroscopicity parameter (κ_m) for ambient aerosol particles**

After transferring the aerosol particles from the filters to the sensor, the sensor was mounted inside the QCM module and the RH was incrementally increased to investigate the water uptake behavior of particles under varying humidity conditions.

We calculated the mass-based hygroscopicity parameter, κ_m, which measures the aerosol particles' ability to take up atmospheric moisture, from the growth factor values obtained from the QCM results (Petters & Kreidenweis, 2007).

\[
\kappa_m = \left( \frac{1}{\alpha_w} - 1 \right) (gf_m - 1) \quad (S2)
\]
where $a_w$ is the water activity and $gf_m$ is the mass-based growth factor of the particle at different RH values. The $a_w$ was approximated as the fractional RH under equilibrium conditions. $\kappa_m$ could be converted to a volume-based hygroscopicity parameter $\kappa_v$, using the following relationship.

$$\kappa_v = \kappa_m \times \frac{\rho_d}{\rho_w}$$

(S4)

where $\rho_d$ is the effective density of the dry particle and $\rho_w$ is the density of water (Mikhailov et al., 2013).

The experiment was repeated for the aerosol particles from different stages of MOUDI to carry out a size-resolved study. In this study, the ambient particles were solid and rigid until they deliquesce at higher RH conditions. The QCM can measure the mass of the particles under different RH conditions till they reach the phase transition point (deliquescence relative humidity, DRH).

**Text S6. Estimation of Deliquescence Relative Humidity (DRH) and Aerosol Liquid Water Content (ALWC)**

The phase transition behavior of ambient aerosols was investigated by analyzing the QCM frequency response curves. During the humidification cycle, the DRH point is identified as the RH at which the oscillation frequencies of QCM response start increasing upon the addition of water (Arenas et al., 2012; Chao et al., 2020). A percentage value, $\Delta f_N$, is obtained by normalizing the oscillation frequency change induced by the aerosol particles at a given RH ($\Delta f_R$) to the oscillation frequency change resulting from a dry sample ($\Delta f_s$ at RH <5%) (Chao et al., 2020).

$$\Delta f_N = \left(\frac{\Delta f_R}{\Delta f_s}\right) \times (-100\%)$$

(S4)

The negative sign indicates that the oscillation frequency of the sensor decreases when RH increases due to water uptake by the aerosol particles. $\Delta f_N$ and its derivative with respect to the measured RH, $d(\Delta f_N)/d(RH)$ were plotted for different size ranges of ambient aerosol samples (Fig. 3). The deliquescence of the particles is indicated by an increase in $\Delta f_N$ or a positive value of $d(\Delta f_N)/d(RH)$. The RH at which $d(\Delta f_N)/d(RH) \geq 1$ is considered as the DRH point for the respective sample (Chao et al., 2020). Following this method, the DRH for (NH$_4$)$_2$SO$_4$ particles was estimated between 76-80% from the QCM measurements (Fig. S4), which agrees with the literature values (Arenas et al., 2012; Martin, 2000; Peng et al., 2022).
Similarly, the DRH values for size-resolved ambient aerosol particles from Munnar were estimated.

Aerosol Liquid Water Content (ALWC) at different RH conditions were also estimated from the size-resolved hygroscopic growth factor values combined with the particle number size distribution (PNSD) measurements, assuming volume conservation during the aerosol hygroscopic process (Bian et al., 2014).

\[
ALWC = \left( \frac{\pi}{6} \sum_i N_i D_{d,i}^3 (g f(D_d, RH) - 1) \right) \rho_w
\]  

(S5)

where \( N_i \) represents the number concentration of dry particles of the \( i \)th bin obtained from SMPS and UV-APS measurements, \( D_{d,i} \) is the particle diameter, and \( \rho_w \) is the density of water.

Text S7. Correction for the RH values inside the QCM module

The flow rate of humidified air in the setup can be impacted by temperature instabilities, potentially affecting the results. To prevent water vapor condensation, it is crucial to maintain the QCM module temperature at a level equal to or higher than the room temperature. To account for any temperature changes inside the QCM module, the RH was corrected using the Clausius-Clapeyron equation.

\[
\ln \frac{P_1}{P_2} = \frac{\Delta H_{\text{vap}}}{R} \left( \frac{1}{T_2} - \frac{1}{T_1} \right)
\]  

(S6)

where \( P_1 \) and \( P_2 \) are the vapor pressures at two temperatures, \( T_1 \) and \( T_2 \), \( \Delta H_{\text{vap}} \) is the enthalpy (heat) of vaporization, and \( R \) is the gas constant (8.3145 J mol\(^{-1}\) K\(^{-1}\)). The actual RH value was determined by computing the ratio of actual vapor pressure to saturation vapor pressure, expressed as a percentage.
Table S1. Mass-based hygroscopic growth factor \((g_{f_m})\) and the corresponding hygroscopicity parameter \((\kappa_m)\) at different RH values in the subsaturated region for different size ranges of ambient aerosol particles (<10 μm) sampled from Munnar using MOUDI. The measurements were obtained from the QCM experiments.

<table>
<thead>
<tr>
<th>Particle size range (μm)</th>
<th>(&lt;0.056)</th>
<th>(0.056-0.1)</th>
<th>(0.1-0.18)</th>
</tr>
</thead>
<tbody>
<tr>
<td>RH (%) (g_{f_m}) (\kappa_m) (\kappa_v) (R) (%) (g_{f_m}) (\kappa_m) (\kappa_v) (R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
</tr>
<tr>
<td>2.87 1.000 0.033 0.040</td>
<td>5.02 1.000 0.054 0.064</td>
<td>3.45 1.000 0.088 0.105</td>
<td>2.93 1.009 0.008 0.010</td>
</tr>
<tr>
<td>11.47 1.004 0.033 0.040</td>
<td>4.03 1.054 0.069 0.082</td>
<td>4.37 1.050 0.062 0.075</td>
<td>27.93 1.009 0.008 0.010</td>
</tr>
<tr>
<td>43.74 1.015 0.020 0.024</td>
<td>64.93 1.145 0.078 0.094</td>
<td>66.04 1.098 0.051 0.061</td>
<td>58.31 1.020 0.014 0.017</td>
</tr>
<tr>
<td>66.73 1.027 0.013 0.016</td>
<td>86.56 1.549 0.085 0.102</td>
<td>86.85 1.567 0.086 0.103</td>
<td>72.95 1.034 0.013 0.015</td>
</tr>
<tr>
<td>80.74 1.051 0.012 0.015</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Particle size range (μm)</th>
<th>(0.18-0.32)</th>
<th>(0.32-0.56)</th>
<th>(0.56-1.0)</th>
</tr>
</thead>
<tbody>
<tr>
<td>RH (%) (g_{f_m}) (\kappa_m) (\kappa_v) (R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
</tr>
<tr>
<td>2.95 1.000 0.071 0.097</td>
<td>3.47 1.000 0.194 0.232</td>
<td>3.09 1.000 0.234 0.256</td>
<td>12.42 1.008 0.059 0.071</td>
</tr>
<tr>
<td>21.84 1.025 0.090 0.109</td>
<td>22.58 1.055 0.188 0.226</td>
<td>21.50 1.056 0.203 0.244</td>
<td>31.36 1.038 0.099 0.109</td>
</tr>
<tr>
<td>47.70 1.055 0.060 0.072</td>
<td>47.33 1.158 0.175 0.211</td>
<td>46.91 1.149 0.169 0.203</td>
<td>55.62 1.071 0.057 0.068</td>
</tr>
<tr>
<td>64.23 1.145 0.081 0.097</td>
<td>64.13 1.365 0.204 0.245</td>
<td>63.93 1.369 0.208 0.250</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Particle size range (μm)</th>
<th>(1.0-1.8)</th>
<th>(1.8-3.2)</th>
<th>(3.2-5.6)</th>
</tr>
</thead>
<tbody>
<tr>
<td>RH (%) (g_{f_m}) (\kappa_m) (\kappa_v) (R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
</tr>
<tr>
<td>3.74 1.000 0.438 0.526</td>
<td>2.80 1.000 0.191 0.229</td>
<td>3.33 1.000 0.097 0.116</td>
<td>12.14 1.061 0.438 0.526</td>
</tr>
<tr>
<td>19.58 1.110 0.451 0.541</td>
<td>21.64 1.053 0.191 0.229</td>
<td>19.70 1.029 0.119 0.143</td>
<td>27.80 1.167 0.435 0.522</td>
</tr>
<tr>
<td>35.00 1.244 0.452 0.543</td>
<td>39.25 1.145 0.225 0.270</td>
<td>35.55 1.056 0.101 0.121</td>
<td>42.85 1.315 0.421 0.505</td>
</tr>
<tr>
<td>47.37 1.411 0.457 0.548</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Particle size range (μm)</th>
<th>(5.6-10.0)</th>
</tr>
</thead>
<tbody>
<tr>
<td>RH (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
<td>(R) (%) (g_{f_m}) (\kappa_m) (\kappa_v)</td>
</tr>
<tr>
<td>3.14 1.000 0.107 0.128</td>
<td>12.25 1.015 0.107 0.128</td>
</tr>
<tr>
<td>21.61 1.030 0.110 0.132</td>
<td>31.22 1.046 0.101 0.121</td>
</tr>
<tr>
<td>39.43 1.067 0.103 0.123</td>
<td>48.00 1.131 0.142 0.171</td>
</tr>
<tr>
<td>56.41 1.177 0.137 0.164</td>
<td>64.77 1.198 0.108 0.129</td>
</tr>
</tbody>
</table>
Table S2. The average mass concentration (μg m\(^{-3}\)) and the mass fraction of the chemical species in NR-PM\(_1\) measured by aerosol chemical speciation monitor (ACSM) during the period from 06.06.2021 to 27.07.2021.

<table>
<thead>
<tr>
<th>NR-PM(_1) Species</th>
<th>Mass concentration (μg m(^{-3}))</th>
<th>Mass fraction</th>
</tr>
</thead>
<tbody>
<tr>
<td>Organics</td>
<td>1.15</td>
<td>0.50</td>
</tr>
<tr>
<td>Sulphate</td>
<td>0.79</td>
<td>0.35</td>
</tr>
<tr>
<td>Nitrate</td>
<td>0.08</td>
<td>0.04</td>
</tr>
<tr>
<td>Ammonium</td>
<td>0.22</td>
<td>0.09</td>
</tr>
<tr>
<td>Chloride</td>
<td>0.04</td>
<td>0.02</td>
</tr>
</tbody>
</table>
Figure S1. Schematic diagram of the experimental setup for providing a wide range of humidity conditions (2-93%) to study the aerosol-water vapor interactions of the samples coated on the quartz crystal microbalance (QCM) sensor mounted inside the module of the chamber platform. Different humidity conditions were created over the aerosol particles by controlling the flowrates of dry and humidified air using the two mass flow controllers (MFCs) connected in parallel. The RH of incoming and outgoing air is measured using two thermohygrometers connected before and after the QCM chamber platform.
Figure S2. The screenshot of the QSoft software showing the oscillation frequency changes of the quartz crystal microbalance (QCM) sensor with time during the humidifying and drying phase of a QCM experiment for the size ranges 56-100 nm and 560 nm-1 μm of ambient aerosol particles from Munnar. (a) shows a decrease in the oscillation frequency change for the particles in the size range 56-100 nm, corresponding to an increase in mass due to water uptake during the humidifying (increasing RH) phase and a further increase in the frequency change when the water content is removed during the drying (decreasing RH) phase. (b) shows an initial decrease in the frequency change due to water uptake by the aerosol particles in the size range 560 nm-1 μm in the lower RH region (<61.2%) during the humidification cycle. The increase in frequency changes and the separation of overtones at RH=61.2% in the humidifying phase indicate that the particles undergo phase transition (deliquescence) at high RH region and become aqueous droplets leading to less viscous and non-rigid particles on the sensor. The initial frequency change value corresponding to the dry particle mass is obtained after the humidification and drying phase, indicating no particle loss. Different curves correspond to different overtone frequencies of the sensor.
Figure S3. Mass-based hygroscopic growth factor $g_{fm}$ of amorphous sucrose particles at different RH values in the subsaturated region obtained from the quartz crystal microbalance (QCM) measurements. The experimental values strictly follow the $g_{fm}$ values retrieved using different models (Norrish, Starzak and Peacock, and Zobrist) from the literature.
Figure S4. The deliquescence phase transition behavior of laboratory-generated \((\text{NH}_4)_2\text{SO}_4\) particles. \(\Delta f_N\) represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the particles at different RH conditions normalized to that of the dry particles at RH < 5%, expressed as percentage. The decrease in the value of \(\Delta f_N\) indicates the water uptake at different RH conditions in the sub-saturated region. The derivative of \(\Delta f_N\) with respect to RH, \(d(\Delta f_N)/d(\text{RH})\) is plotted against different RH conditions and the respective RH values for \(d(\Delta f_N)/d(\text{RH}) \geq 1\), marked by the black dotted line, represent the deliquescence relative humidity (DRH) range (76-80%) for the \((\text{NH}_4)_2\text{SO}_4\) particles.
Figure S5. Aerosol Liquid Water Content (ALWC) in size-resolved ambient aerosol particles from Munnar across varied relative humidity (RH) conditions in the subsaturated regime. ALWC under different RH conditions is estimated by integrating size-resolved hygroscopic growth factor values with particle number size distribution (PNSD) measurements, assuming volume conservation during aerosol water uptake processes. The marker sizes correspond linearly to particle size, while marker colors signify the mean value of the mass-based hygroscopicity parameter ($\kappa_m$) for each size-range, indicated by the color palette. The total ALWC averaged over all particles below 10 μm in size is also shown in inset.
References:


