Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique

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Abstract

Comprehending the intricate interplay between atmospheric aerosols and water vapour in subsaturated regions is vital for accurate modelling of aerosol–cloud–radiation–climate dynamics. But the microphysical mechanisms governing these interactions with ambient aerosols remain inadequately understood. Here we report results from high-altitude, relatively pristine site in Western-Ghats of India during monsoon, serving as a baseline for climate processes in one of the world's most polluted regions. Utilizing a novel quartz crystal microbalance (QCM) approach, we conducted size-resolved sampling to analyse humidity-dependent growth factors, hygroscopicity, deliquescence behaviour, and aerosol liquid water content (ALWC). Fine-mode aerosols ([?]2.5 μ m) exhibited size-dependent interactions with water vapour, contributing significantly to ALWC. Deliquescence was observed in larger aerosols (>180 nm), influenced by organic species, with deliquescence relative humidity (DRH) lower than that of pure inorganic salts. This research highlights the significance of understanding ambient aerosol-water interactions and hygroscopicity for refining climate models in subsaturated conditions.

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- 25

26 **KEYPOINTS**

- QCM technique revealed aerosol properties at low RH values offering insights that
 may not be captured by traditional size-based measurements
- The size-dependent variations in aerosol properties below DRH emphasize the need to
 understand how aerosol properties change with RH history
- Organic in aerosols complicate its cloud forming ability, necessitating additional
 studies in diverse environments to improve climate models

33 ABSTRACT

Comprehending the intricate interplay between atmospheric aerosols and water vapour in 34 subsaturated regions is vital for accurate modelling of aerosol-cloud-radiation-climate 35 dynamics. But the microphysical mechanisms governing these interactions with ambient 36 aerosols remain inadequately understood. Here we report results from high-altitude, relatively 37 pristine site in Western-Ghats of India during monsoon, serving as a baseline for climate 38 processes in one of the world's most polluted regions. Utilizing a novel quartz crystal 39 microbalance (QCM) approach, we conducted size-resolved sampling to analyse humidity-40 dependent growth factors, hygroscopicity, deliquescence behaviour, and aerosol liquid water 41 content (ALWC). Fine-mode aerosols ($\leq 2.5 \mu m$) exhibited size-dependent interactions with 42 water vapour, contributing significantly to ALWC. Deliquescence was observed in larger 43 aerosols (>180 nm), influenced by organic species, with deliquescence relative humidity 44 (DRH) lower than that of pure inorganic salts. This research highlights the significance of 45 understanding ambient aerosol-water interactions and hygroscopicity for refining climate 46 models in subsaturated conditions. 47

48 PLAIN LANGUAGE SUMMARY

Aerosol particles interact with water vapour in the atmosphere. Understanding these 49 interactions in sub – and super-saturated regions is crucial because they affect processes such 50 as cloud formation, radiation, and climate. We collected ambient aerosol samples from 51 Western Ghats in India to understand natural processes that happen without significant 52 human interventions. Using a special technique involving a quartz crystal sensor, we 53 measured different aspects, such as how aerosols grow with humidity, their ability to uptake 54 the water molecules, and when they transition from solid to liquid (deliquescence). 55 Interestingly, fine particles, especially those smaller than 2.5 µm, exhibited unique 56 57 behaviours. While larger particles underwent a transition from solid to liquid under certain conditions, this didn't happen for the smaller particles. These findings highlight the 58 importance of understanding these interactions for more accurate climate predictions. 59

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60 1. INTRODUCTION

61 Understanding the hygroscopicity of atmospheric aerosols is crucial for assessing cloud

- 62 formation and their climate and air pollution impact (Cheung et al., 2015). Investigation of
- 63 the hygroscopicity of ambient aerosols poses significant challenges due to their chemical
- 64 complexity, particle size variation, phase state, and viscosity. Techniques like Fourier
- transform infrared spectroscopy (FTIR) (Y. Liu et al., 2008; Y. Liu & Laskin, 2009), quartz
- crystal microbalance (QCM) (Chao et al., 2020; Demou et al., 2003; P. Liu et al., 2016a; P.
- Liu, Song, et al., 2018a), Raman spectroscopy (Ling & Chan, 2008; Y. J. Liu et al., 2008),
- electrodynamic balance (EDB) (Choi & Chan, 2002; Peng & Chan, 2001; Pope et al., 2010),
- optical microscopy (OM) (Ahn et al., 2010; Eom et al., 2014; Gupta et al., 2015),
- 70 hygroscopicity tandem differential mobility analysis (HTDMA) (Cheung et al., 2015; Prenni
- et al., 2007; Zieger et al., 2017a) and size-selected cloud condensation nuclei (CCN)
- spectrometry (Petters et al., 2007; Pöhlker et al., 2016; Rose et al., 2008) have been employed
- to study the hygroscopicity of laboratory generated and ambient aerosols (Tang et al.,
- 74 2019). However, a consensus on the most effective method is lacking, leading to inconsistent
- and incomparable results. Hygroscopicity measurements primarily focus on accumulation
- 76 mode particles, limiting data on nucleation and coarse mode particles with complex
- behaviours. Discrepancies in hygroscopic properties and organic and inorganic species
- 78 composition in atmospheric aerosols result in diverse growth factors and phase transitions,
- rucial for accurate climate modelling (Li et al., 2021).
- 80

The QCM is effective in determining mass–based hygroscopicity and physical property
variations of atmospherically–relevant aerosols (Demou et al., 2003; P. Liu et al., 2016b; P.

- Liu, Song, et al., 2018b). It overcomes the limitations of traditional methods like HTDMA,
- 84 which has a narrow particle size range with limited relative humidity (RH) resolution (Zhao
- et al., 2022). Conventional HTDMA methods inadequately reveal hygroscopicity
- 86 heterogeneity and phase transitions in wider aerosol size range, impacting processes like
- 87 CCN activation and aerosol liquid water (Li et al., 2021). QCM's mass-based measurements
- 88 accurately quantify aerosol hygroscopicity behaviour across a wide RH range in the
- 89 atmosphere, detailing physical property variations during phase transitions and help to
- 90 delineate water uptake mechanisms, including solubility and water diffusion limitations (P.
- 91 Liu, Li, et al., 2018; P. Liu, Song, et al., 2018b).
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India's climate is distinct and intricate, and as per the 2020 climate change assessment report, 93 94 India has faced challenges like temperature rise and extreme weather events since the mid-20th century. A key hurdle in addressing these issues is the absence of systematic aerosol 95 characteristic measurements, particularly in aerosol-water vapor interactions in the 96 97 subsaturated regime (Cheung et al., 2015). Aerosol particles from a forest-surrounded by pristine area are anticipated to contain substantial biogenic organic aerosols. The dynamic 98 99 chemical transformation of secondary organic aerosols (SOA) in the atmosphere leads to intricate compositions. Atmospheric processing further complicates the understanding of 100 101 phase state and properties of atmospheric aerosols (Rastak et al., 2017). In this study, QCM measurements are utilized to explore aerosol hygroscopicity dependence on RH, chemical 102 composition, and size across a broad range of ambient aerosols in India's relatively pristine 103 104 setting.

105 2. MATERIALS AND METHODS

Size-resolved ambient aerosols were sampled at the Natural Aerosol and Bioaerosol High 106 Altitude (NABHA) Laboratory at the College of Engineering, Munnar (10.0930° N, 77.0682° 107 E; 1600 m above mean sea level), a high-altitude and typically clean site situated in the 108 Western Ghats of India, using a 10-stage micro-orifice uniform deposit impactor (MOUDI-II 109 120-R, TSI) (V. Marple et al., 2014; V. A. Marple et al., 1991) during the monsoon season 110 (August-September) of 2021. Ambient particles collected on PTFE (Polytetrafluoroethylene) 111 112 filters were transferred to a hydrophobic SiO₂-coated quartz sensor by gently pressing the filter paper onto the sensor. The hygroscopic growth factor, hygroscopicity parameter and 113 deliquescence relative humidity (DRH) were estimated using a highly sensitive mass balance 114 115 instrument, QCM (QSense Analyzer, Biolin Scientific) (P. Liu, Song, et al., 2018a; Reviakine et al., 2011). The accuracy and robustness of the method were demonstrated by performing 116 measurements of the hygroscopic growth factor and DRH for sucrose and (NH₄)₂SO₄ 117 particles, respectively, which were compared with previously reported results (Arenas et al., 118 2012a; Chao et al., 2020; Martin, 2000; Norrish, 1966; Peng et al., 2022; Starzak & Peacock, 119 1997; Zobrist et al., 2011) (Fig. S3 and S4). The hygroscopic growth factor and the 120 121 corresponding hygroscopicity parameter, κ , were determined over a wide range of RH conditions using ĸ-Köhler theory (Petters & Kreidenweis, 2007), where κ represents a 122 quantitative measure of aerosol water uptake characteristics and CCN activity (Zhao et al., 123 2022). The water uptake characteristics and the phase transition behaviours of the ambient 124

samples were investigated and compared between different size ranges of particles below 10 um collected using the MOUDI sampler. The Aerosol Liquid Water Content (ALWC) (Bian et al., 2014) at different RH conditions was also estimated from corresponding hygroscopicity parameters for each size range of particles (Fig. S5). The number size distributions of the ambient aerosol particles of size 10 to 420 nm [measured in parallel using a scanning mobility particle sizer – SMPS, comprised of an electrostatic classifier (EC, TSI Model 3082) equipped with a differential mobility analyser (DMA, TSI Model 3081) and a condensation particle counter (CPC, TSI Model 3750)], were then converted to mass size distributions assuming a particle density of 1.2 g cm⁻³ (DeCarlo et al., 2004). The supermicron particle mass size distribution for the same season, but a different year, was used in this study, which had been obtained using an Ultraviolet Aerodynamic Particle Sizer (UV-APS, TSI Inc., model 3314) (Valsan et al., 2016).

Meteorological parameters were recorded using an automatic weather station (AWS, Clima Sensor US) during the sampling period. Figure 1 shows the sampling site – Munnar, along with aerosol optical depth (AOD) over India during the monsoon season, indicating the relatively pristine nature of the sampling site. During the measurement, airmasses predominantly arrived from the southwest direction, bringing clean marine influx to the observational site and resulting in a low influence of anthropogenic emissions (Fig. 1). A more detailed description of the instruments, experimental techniques, and estimation of various parameters is provided in the Supporting Information (SI).



Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate 166 167 Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon 168 season of 2021 (June - September). The AOD distribution clearly indicates a relatively low aerosol loading over 169 the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind 170 rose diagram shown in the inset shows the average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August - September 2021). The prevailing air masses mostly 171 172 originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the 173 observational site, confirming the relatively low influence of anthropogenic activities.

174

175 **3. RESULTS AND DISCUSSION**

176 The QCM sensor's oscillation frequency variation (Δf) signifies water uptake and release,

177 providing insights into adsorption, desorption, and physical states of particles during solid-to-

aqueous phase transition (Arenas et al., 2012b). Normalizing the frequency shift at higher RH

to that of deposited dry sample (RH <5%) yields the percentage value, Δf_N (Chao et al.,

180 2020). Figure 2 shows Δf_N for the ambient aerosol particles across various RH levels and

sampled size ranges. A negative sign in Δf_N indicates frequency reduction with increasing RH

due to water uptake. Further details of Δf_N and its derivative, $d(\Delta f_N)/(RH)$, are available in SI.

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184 For the particle size <180 nm (Fig. 2a), Δf_N decreased, implying water uptake, with <56 nm

particles exhibiting the lowest value. In Figure 2b and 2c, particles showed increased water

uptake at lower RH, evidenced by Δf_N decrease. However, Δf_N suddenly increased at specific

187 RH, indicating a drastic water uptake leading to deliquescence-induced phase transition.

188 Figures 2a-c revealed deliquescence in particles >180 nm at different RH (known as DRH),

absent in those <180 nm. Past studies reported no deliquescence for ambient and laboratory-

190 generated atmospherically relevant particles even at high RH, attributing them to organic

191 nature (Arenas et al., 2012a; Brooks et al., 2002; Chao et al., 2020; Peng et al., 2022). We

192 hypothesize that particles <180 nm were likely dominated by freshly formed secondary

193 organic aerosols (SOA) from biogenic volatile organic compound (VOC) oxidation,

supported by the densely vegetated/forest region and season. Concurrent quasi-continuous

measurements during the same campaign indicated upto $\sim 90\%$ organic fraction in NR-PM₁

196 (non-refractory particulate matter with an aerodynamic diameter $\leq 1 \mu m$; Tab. S2) with details

197 discussed in subsequent studies.



199 Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For 200 panels a, b, and c, Δf_N represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) 201 conditions normalized to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in 202 the value of Δf_N for each size range for the sampled ambient aerosol particles indicates the water uptake at 203 different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size 204 205 ranges. In panels d, e, and f, the derivative of Δf_N with respect to RH (d(Δf_N)/d(RH)) is plotted against RH to 206 determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. 207 The RH values at which $d(\Delta f_N)/d(RH)$ becomes ≥ 1 (marked by the dotted line) represent the DRH values for the 208 individual aerosol size ranges.

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Following Chao et al. (2020), DRH, indicated by $d(\Delta f_N)/(RH) \ge 1$ ranged from ~60 – 68% for 210 three size ranges (180 nm - 320 nm; 320 nm - 560 nm; 560 nm -1μ m) in this study (Fig. 211 2e). Above 180 nm, inorganic salts, like (NH₄)₂SO₄, may contribute to DRH appearance (Hu 212 et al., 2010) (Fig. 2b,e). These DRH values, smaller than pure (NH₄)₂SO₄ (DRH=80%), 213 suggest mixed salts and/or organic compounds in the sampled aerosol particles. The 214 deliquescence behaviour of organic-inorganic mixtures introduces complexity due to 215 solubility limitations, influenced by the organic species in the aerosol particles. Previous 216 studies noted reduced DRH of inorganic species and sometimes, unaffected by organics 217 (Smith et al., 2011, 2012, 2013). Thus, we hypothesize 180 nm to 1 µm particles at this site 218 reflect a complex mixture of inorganic salts like (NH₄)₂SO₄ and water-soluble organic 219 compounds. Supermicron particles (>1 μ m), except 1 – 1.8 μ m, showed <17% $\Delta f_{\rm N}$ indicating 220 221 lower water uptake (Fig. 2c). Beyond maximum water uptake, the three size ranges exhibited 222 different DRH values. $1.8 - 3.2 \mu m$ and $5.6 - 10 \mu m$ exhibited DRH between 56 - 68%, suggesting a mixture of water-soluble organic compounds and inorganic salts. However, 3.2 223 $-5.6 \,\mu\text{m}$ showed a very high DRH (>81%) with the organic fraction minimally affecting 224

DRH due to solubility limitations. For $1.8 - 3.2 \mu m$ and $5.6 - 10 \mu m$, reduced DRH

compared to pure inorganic salts suggests a mix of water-soluble organic and inorganic

- 227 compounds (Smith et al., 2011, 2012, 2013). The $1.0 1.8 \mu m$ range exhibited a >30% Δf_N
- reduction at ~42% RH, indicating a higher water uptake and the observed DRH appeared to
- be ~45 53%, consistent with the highest κ_m (~0.44, see discussion below) among all
- 230 measured size ranges (Chao et al., 2020). The low DRH at this site suggests presence of
- highly water-soluble organic material within the organic-inorganic (sea salt) mixture in this
- size range.
- 233
- Figure 3 shows the size-resolved mass-based hygroscopic growth factors (gf_m) and the
- corresponding mass-based hygroscopicity parameters (κ_m) of ambient aerosol particles
- calculated from QCM measurements (Text S4 and S5) over a wide RH range (2 90%). In
- Figure 3a, the growth factor increases with water uptake in humid air, showing size-
- 238 dependent variations across different RH levels (indicated by markers). Each size range had a
- 239 maximum RH limit beyond which gf_m estimation was impossible due to deliquescence, as
- observed in previous studies (Chao et al., 2020). Accordingly, the highest $gf_m=1.72$ was
- recorded at RH=90% for the 100 180 nm particles. The pronounced variations in gf_m
- 242 between different-sized particles at a given RH indicate varying chemical composition and
- 243 hygroscopicity across the sampled sizes.
- 244

Figure 3b shows the size-dependent average κ_m values, calculated from the gf_m values in 245 Figure 3a. Each particle size range exhibited minimal κ_m variation (average $\kappa_m \pm 0.007$), 246 suggesting ideal solution behaviour across the wide RH range (Z. Wang et al., 2017). The 247 248 average κ_m values (indicated by dashed lines in Figure 3b) of the size-resolved ambient particles varied significantly between 0.016 (< 0.056 nm) and 0.44 ($1.0 - 1.8 \mu$ m) over the 249 investigated RH range. This implies κ_m as a function of size, and indicates distinct chemical 250 compositions for each size fraction. For example, the small increase in κ_m from 0.016 to 251 0.078 for particles <56 nm to 56-100 nm, respectively, is potentially owing to aging 252 processes (Zhang et al., 2023). Aerosol particles <320 nm exhibited very low average κ_m 253 (0.06), which may be primarily due to the strong dominance of organic compounds in these 254 255 size ranges (Demou et al., 2003; Li et al., 2021; Shi et al., 2022). The particles in the size range of 320 nm $- 1 \mu m$ were moderately hygroscopic (average $\kappa_m = 0.19$) indicating the 256 influence of inorganic salts (Wu et al., 2016). The particles between $1.0 - 1.8 \mu m$ exhibited 257 highest κ_m (0.44), which implies the presence of highly hygroscopic material such as sea salt 258

- (Zieger et al., 2017b) mixed with organic compounds. Even larger particles exhibited reduced 259
- κ_m , possibly due to the presence of dust particles (Koehler et al., 2009), transported at this site 260
- during monsoon season (Valsan et al., 2016). The κ -Köhler parameterization fit accurately 261
- elucidates the gfm within each size range over a wide range of RH values (Petters & 262
- Kreidenweis, 2007). 263
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292 Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, 293 Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor (gfm) 294 derived using a quartz crystal microbalance (OCM) for ten different size ranges of ambient aerosol particles at 295 different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the 296 corresponding κ-Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity 297 parameters, κ_m . The error bars represent the variations in gfm averaged over the mass change corresponding to 298 different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the 299 mean κ_m values corresponding to the respective size ranges. (b) The data points are the κ_m values calculated 300 based on the gf_m (as shown in (a)) using the κ -Köhler theory for different RH conditions in the subsaturated 301 regime. The dotted lines represent the mean κ_m value for each size range of ambient aerosol particles and the 302 error bars represent one standard deviation.

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Figure 4 shows the size dependency of κ_m in the subsaturated regime measured by QCM 304 together with $\kappa_{\rm CCN}$ (Text S2) in the supersaturated regime determined by size-resolved CCN 305 measurements and aerosol size distributions obtained by SMPS and UV-APS. The submicron 306 particles show a bimodal number size distribution with an Aitken mode peak at 63 nm and an 307 accumulation mode at 145 nm, while the mass size distributions peak at 350 nm and 3 μ m, 308

respectively. The average κ_{CCN} increased from a value of 0.28 for the Aitken mode particles 309 to 0.47 for the accumulation mode particles. In the submicron region, κ_m exhibited a nominal 310 increase with particle size (Wu et al., 2016). in parallel with the increase in κ_{CCN} , albeit at 311 much lower absolute values. This increase likely reflects the decrease of organic fraction with 312 size in the submicron range, which is supported by the appearance of deliquescence in the 313 size range above 180 nm, indicating the presence of inorganic salts. In the supermicron 314 region (for particles $\geq 1 \,\mu m$), κ_m peaked due to the potential presence of NaCl particles in the 315 size range $1.0 - 1.8 \,\mu\text{m}$, and decreased again in the higher size ranges (>1.8 μm) likely 316 317 because of the presence of dust particles. The average κ_m value (0.18) obtained from QCM measurements across all the sampled size ranges in the subsaturated regime was lower than 318 319 the average κ_{CCN} (0.39), likely due to the solubility limitation (Dusek et al., 2011; Hersey et al., 2013; Rastak et al., 2017; Riipinen et al., 2015; Wittbom et al., 2018) of organic and 320 inorganic compounds in ambient aerosol particles below the DRH. Most inorganic species 321 are completely dissolved beyond the DRH point in the subsaturated regime, allowing κ_{CCN} to 322 assume complete particle solubility (Pajunoja et al., 2015; Petters & Kreidenweis, 2007). 323



340 Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance 341 (QCM) experiments (κ_m), and from size-resolved cloud condensation nuclei (CCN) measurements (κ_{CCN}). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 -342 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass 343 344 size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by 345 assuming a density of 1.2 g cm⁻³ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements 346 347 during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived 348 from QCM experiments (κ_{m} ; orange points) and size-resolved CCN measurements (κ_{CCN} ; blue points) are shown 349 for the comparison. The error bars for κ_m and κ_{CCN} indicate the measurement uncertainty and variability, 350 respectively.

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352 Unlike QCM measurements, the hygroscopicity parameters obtained in previous studies

using the HTDMA technique (κ_{HTDMA}) are often consistent with κ_{CCN} , as the particle

hygroscopicity is determined by particle size increase above DRH point, unaffected by 354 solubility limits (Pajunoja et al., 2015; Y. Wang et al., 2018; Wu et al., 2013). The technical 355 limitations, particularly limited diameter changes corresponding to gfm values less than 2.0, 356 prevent HTDMA from measuring aerosol hygroscopic properties below the DRH point in the 357 subsaturated regime (Laskina et al., 2015). In contrast, the QCM technique, which is not 358 subject to these limitations, can be used to understand the water uptake characteristics of 359 aerosol particles in the subsaturated regime below DRH and substantially enhance our 360 understanding about solubility limitations of organic and inorganic compounds in the lower 361 362 RH region. Such an enhancement in the knowledge is crucial for accurately quantifying the radiative forcing effects of ambient aerosol particles in the atmosphere. The strong size 363 dependence of κ_m also indicated varying chemical composition for different size ranges 364 365 measured in this study.

366 4. SUMMARY AND ATMOSPHERIC IMPLICATIONS

We report the first results based on a high-sensitivity QCM technique to investigate the mass-367 368 based growth factor and hygroscopicity parameter of size-resolved ambient aerosols over a wide range of RH from a relatively pristine high-altitude site in India. For the investigated 369 size ranges, the pronounced variations in gfm below the DRH provided an opportunity to 370 better understand the changes in ambient aerosol properties even at low RH values, which 371 may not be revealed by size-based growth factor measurements (Hu et al., 2010; Laskina et 372 al., 2015). These measurements clearly highlight the important and critical role in knowing 373 the changes in aerosol properties based on their RH history for an improved understanding of 374 water uptake, phase transition, and radiative impact of atmospheric aerosol particles (Zhao et 375 al., 2022). Based on the estimated κ_m values, the ambient aerosol particles exhibited the 376 behaviour of an ideal solution (Pajunoja et al., 2015) and strong size-dependent chemical 377 composition. We further observed a pronounced size dependency of the DRH values 378 potentially resulting from complexities of organic solubility in organic-inorganic mixtures 379 (Li et al., 2021). Such a complexity owing to the presence of various organic species may 380 381 alter the CCN behaviour of atmospheric aerosol particles due to phase transitions and changes in physical properties. The understanding of CCN activation in the supersaturated 382 regime under the assumption of complete solubility of particles is relatively well established. 383 But the understanding of the thermodynamic properties associated with particle growth in the 384 385 subsaturated regime still remains a challenge. Our findings emphasize the need for additional experiments on ambient and atmospherically relevant laboratory-generated aerosol particles 386

- using high-sensitivity techniques like QCM. We demonstrated the importance of this
- technique to better understand the rapid changes in aerosol properties resulting from exposure
- to a wide range of atmospheric RH conditions. The enhancement in our understanding of the
- 390 complex interplay between water vapour and aerosol particles will help in developing more
- accurate models to effectively describe the role of aerosols in atmospheric processes to
- reduce climate uncertainties and assess the impact of air pollution on human and ecosystem
- 393 health.

394 ASSOCIATED CONTENT

395

396 Author contributions

SSG conceived the idea. SSG and PL conceptualized the study and designed the research. CJ 397 further developed and validated the QCM measurements for the ambient aerosol studies. CJ 398 performed the field measurement campaign to collect the aerosol samples with support from 399 AS, KNK, and RKA. CJ performed all the laboratory experiments using QCM with support 400 from SS. GVJ performed the satellite data analysis to obtain the AOD values. CJ carried out 401 402 the data analysis obtained from QCM with input from PL and SSG. CJ and AS performed the scientific interpretation of the QCM data under the mentorship of RR, SSG and PL. CJ wrote 403 the first draft of manuscript under the mentorship of SSG with inputs from PL and RR, and 404 further edits from SY. MOA and STM further provided critical and valuable inputs on the 405 manuscript. 406

407

408 **Open research**

- 409 The data used in the manuscript has been deposited in an open research repository as Excel
- 410 files, accessible at https://doi.org/10.6084/m9.figshare.24512377. MODIS data were sourced
- 411 from the Level 2 and Atmosphere Archive and Distribution System (LAADS)
- 412 (https://doi.org/10.5067/MODIS/MYD04_3K.061). Figures were prepared using Igor Pro
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- 415

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Figure.



Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon season of 2021 (June - September). The AOD distribution clearly indicates a relatively lower aerosol loading over the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind rose diagram shown in the inset is average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August – September 2021). The prevailing air masses mostly originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the observational site, confirming the relatively low influence of anthropogenic activities.



Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For panels a, b, and c, Δf_N represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) conditions normalised to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in the value of Δf_N for each size range for the sampled ambient aerosol particles indicates the water uptake at different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size ranges. In panels d, e, and f the derivative of Δf_N with respect to RH (d(Δf_N)/d(RH)) is plotted against RH to determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. The RH values at which d(Δf_N)/d(RH) becomes ≥ 1 (marked by the dotted line) represent the DRH values for the individual aerosol size ranges.



Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor (gf_m) derived using a quartz crystal microbalance (QCM) for ten different size ranges of ambient aerosol particles at different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the corresponding κ -Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity parameter, κ_m . The error bars represent the variations in gf_m averaged over the mass change corresponding to different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the mean κ_m values corresponding to respective size ranges. (b) The data points are the κ_m values calculated based on the gf_m (as shown in (a)) using the κ -Köhler theory for different RH conditions in the subsaturated regime. The dotted lines represent the mean κ_m value for each size range of ambient aerosol particles and the error bars represent one standard deviation.



Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance (QCM) experiments (κ_m), and from size-resolved cloud condensation nuclei (CCN) measurements (κ_{CCN}). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 – 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by assuming a density of 1.2 g cm⁻³ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived from QCM experiments (κ_m ; orange points) and size-resolved CCN measurements (κ_{CCN} ; blue points) are shown for the comparison. The error bars for κ_m and κ_{CCN} indicate the measurement uncertainty and variability, respectively.

Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique

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26 **KEYPOINTS**

- QCM technique revealed aerosol properties at low RH values offering insights that
 may not be captured by traditional size-based measurements
- The size-dependent variations in aerosol properties below DRH emphasize the need to
 understand how aerosol properties change with RH history
- Organic in aerosols complicate its cloud forming ability, necessitating additional
 studies in diverse environments to improve climate models

33 ABSTRACT

Comprehending the intricate interplay between atmospheric aerosols and water vapour in 34 subsaturated regions is vital for accurate modelling of aerosol-cloud-radiation-climate 35 dynamics. But the microphysical mechanisms governing these interactions with ambient 36 aerosols remain inadequately understood. Here we report results from high-altitude, relatively 37 pristine site in Western-Ghats of India during monsoon, serving as a baseline for climate 38 processes in one of the world's most polluted regions. Utilizing a novel quartz crystal 39 microbalance (QCM) approach, we conducted size-resolved sampling to analyse humidity-40 dependent growth factors, hygroscopicity, deliquescence behaviour, and aerosol liquid water 41 content (ALWC). Fine-mode aerosols ($\leq 2.5 \mu m$) exhibited size-dependent interactions with 42 water vapour, contributing significantly to ALWC. Deliquescence was observed in larger 43 aerosols (>180 nm), influenced by organic species, with deliquescence relative humidity 44 (DRH) lower than that of pure inorganic salts. This research highlights the significance of 45 understanding ambient aerosol-water interactions and hygroscopicity for refining climate 46 models in subsaturated conditions. 47

48 PLAIN LANGUAGE SUMMARY

Aerosol particles interact with water vapour in the atmosphere. Understanding these 49 interactions in sub – and super-saturated regions is crucial because they affect processes such 50 as cloud formation, radiation, and climate. We collected ambient aerosol samples from 51 Western Ghats in India to understand natural processes that happen without significant 52 human interventions. Using a special technique involving a quartz crystal sensor, we 53 measured different aspects, such as how aerosols grow with humidity, their ability to uptake 54 the water molecules, and when they transition from solid to liquid (deliquescence). 55 Interestingly, fine particles, especially those smaller than 2.5 µm, exhibited unique 56 57 behaviours. While larger particles underwent a transition from solid to liquid under certain conditions, this didn't happen for the smaller particles. These findings highlight the 58 importance of understanding these interactions for more accurate climate predictions. 59

2

60 1. INTRODUCTION

61 Understanding the hygroscopicity of atmospheric aerosols is crucial for assessing cloud

- 62 formation and their climate and air pollution impact (Cheung et al., 2015). Investigation of
- 63 the hygroscopicity of ambient aerosols poses significant challenges due to their chemical
- 64 complexity, particle size variation, phase state, and viscosity. Techniques like Fourier
- transform infrared spectroscopy (FTIR) (Y. Liu et al., 2008; Y. Liu & Laskin, 2009), quartz
- crystal microbalance (QCM) (Chao et al., 2020; Demou et al., 2003; P. Liu et al., 2016a; P.
- Liu, Song, et al., 2018a), Raman spectroscopy (Ling & Chan, 2008; Y. J. Liu et al., 2008),
- electrodynamic balance (EDB) (Choi & Chan, 2002; Peng & Chan, 2001; Pope et al., 2010),
- optical microscopy (OM) (Ahn et al., 2010; Eom et al., 2014; Gupta et al., 2015),
- 70 hygroscopicity tandem differential mobility analysis (HTDMA) (Cheung et al., 2015; Prenni
- et al., 2007; Zieger et al., 2017a) and size-selected cloud condensation nuclei (CCN)
- spectrometry (Petters et al., 2007; Pöhlker et al., 2016; Rose et al., 2008) have been employed
- to study the hygroscopicity of laboratory generated and ambient aerosols (Tang et al.,
- 74 2019). However, a consensus on the most effective method is lacking, leading to inconsistent
- and incomparable results. Hygroscopicity measurements primarily focus on accumulation
- 76 mode particles, limiting data on nucleation and coarse mode particles with complex
- behaviours. Discrepancies in hygroscopic properties and organic and inorganic species
- 78 composition in atmospheric aerosols result in diverse growth factors and phase transitions,
- rucial for accurate climate modelling (Li et al., 2021).
- 80

The QCM is effective in determining mass–based hygroscopicity and physical property
variations of atmospherically–relevant aerosols (Demou et al., 2003; P. Liu et al., 2016b; P.

- Liu, Song, et al., 2018b). It overcomes the limitations of traditional methods like HTDMA,
- 84 which has a narrow particle size range with limited relative humidity (RH) resolution (Zhao
- et al., 2022). Conventional HTDMA methods inadequately reveal hygroscopicity
- 86 heterogeneity and phase transitions in wider aerosol size range, impacting processes like
- 87 CCN activation and aerosol liquid water (Li et al., 2021). QCM's mass-based measurements
- 88 accurately quantify aerosol hygroscopicity behaviour across a wide RH range in the
- 89 atmosphere, detailing physical property variations during phase transitions and help to
- 90 delineate water uptake mechanisms, including solubility and water diffusion limitations (P.
- 91 Liu, Li, et al., 2018; P. Liu, Song, et al., 2018b).
- 92

3

India's climate is distinct and intricate, and as per the 2020 climate change assessment report, 93 94 India has faced challenges like temperature rise and extreme weather events since the mid-20th century. A key hurdle in addressing these issues is the absence of systematic aerosol 95 characteristic measurements, particularly in aerosol-water vapor interactions in the 96 97 subsaturated regime (Cheung et al., 2015). Aerosol particles from a forest-surrounded by pristine area are anticipated to contain substantial biogenic organic aerosols. The dynamic 98 99 chemical transformation of secondary organic aerosols (SOA) in the atmosphere leads to intricate compositions. Atmospheric processing further complicates the understanding of 100 101 phase state and properties of atmospheric aerosols (Rastak et al., 2017). In this study, QCM measurements are utilized to explore aerosol hygroscopicity dependence on RH, chemical 102 composition, and size across a broad range of ambient aerosols in India's relatively pristine 103 104 setting.

105 2. MATERIALS AND METHODS

Size-resolved ambient aerosols were sampled at the Natural Aerosol and Bioaerosol High 106 Altitude (NABHA) Laboratory at the College of Engineering, Munnar (10.0930° N, 77.0682° 107 E; 1600 m above mean sea level), a high-altitude and typically clean site situated in the 108 Western Ghats of India, using a 10-stage micro-orifice uniform deposit impactor (MOUDI-II 109 120-R, TSI) (V. Marple et al., 2014; V. A. Marple et al., 1991) during the monsoon season 110 (August-September) of 2021. Ambient particles collected on PTFE (Polytetrafluoroethylene) 111 112 filters were transferred to a hydrophobic SiO₂-coated quartz sensor by gently pressing the filter paper onto the sensor. The hygroscopic growth factor, hygroscopicity parameter and 113 deliquescence relative humidity (DRH) were estimated using a highly sensitive mass balance 114 115 instrument, QCM (QSense Analyzer, Biolin Scientific) (P. Liu, Song, et al., 2018a; Reviakine et al., 2011). The accuracy and robustness of the method were demonstrated by performing 116 measurements of the hygroscopic growth factor and DRH for sucrose and (NH₄)₂SO₄ 117 particles, respectively, which were compared with previously reported results (Arenas et al., 118 2012a; Chao et al., 2020; Martin, 2000; Norrish, 1966; Peng et al., 2022; Starzak & Peacock, 119 1997; Zobrist et al., 2011) (Fig. S3 and S4). The hygroscopic growth factor and the 120 121 corresponding hygroscopicity parameter, κ , were determined over a wide range of RH conditions using ĸ-Köhler theory (Petters & Kreidenweis, 2007), where κ represents a 122 quantitative measure of aerosol water uptake characteristics and CCN activity (Zhao et al., 123 2022). The water uptake characteristics and the phase transition behaviours of the ambient 124

samples were investigated and compared between different size ranges of particles below 10 um collected using the MOUDI sampler. The Aerosol Liquid Water Content (ALWC) (Bian et al., 2014) at different RH conditions was also estimated from corresponding hygroscopicity parameters for each size range of particles (Fig. S5). The number size distributions of the ambient aerosol particles of size 10 to 420 nm [measured in parallel using a scanning mobility particle sizer – SMPS, comprised of an electrostatic classifier (EC, TSI Model 3082) equipped with a differential mobility analyser (DMA, TSI Model 3081) and a condensation particle counter (CPC, TSI Model 3750)], were then converted to mass size distributions assuming a particle density of 1.2 g cm⁻³ (DeCarlo et al., 2004). The supermicron particle mass size distribution for the same season, but a different year, was used in this study, which had been obtained using an Ultraviolet Aerodynamic Particle Sizer (UV-APS, TSI Inc., model 3314) (Valsan et al., 2016).

Meteorological parameters were recorded using an automatic weather station (AWS, Clima Sensor US) during the sampling period. Figure 1 shows the sampling site – Munnar, along with aerosol optical depth (AOD) over India during the monsoon season, indicating the relatively pristine nature of the sampling site. During the measurement, airmasses predominantly arrived from the southwest direction, bringing clean marine influx to the observational site and resulting in a low influence of anthropogenic emissions (Fig. 1). A more detailed description of the instruments, experimental techniques, and estimation of various parameters is provided in the Supporting Information (SI).



Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate 166 167 Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon 168 season of 2021 (June - September). The AOD distribution clearly indicates a relatively low aerosol loading over 169 the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind 170 rose diagram shown in the inset shows the average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August - September 2021). The prevailing air masses mostly 171 172 originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the 173 observational site, confirming the relatively low influence of anthropogenic activities.

174

175 **3. RESULTS AND DISCUSSION**

176 The QCM sensor's oscillation frequency variation (Δf) signifies water uptake and release,

177 providing insights into adsorption, desorption, and physical states of particles during solid-to-

aqueous phase transition (Arenas et al., 2012b). Normalizing the frequency shift at higher RH

to that of deposited dry sample (RH <5%) yields the percentage value, Δf_N (Chao et al.,

180 2020). Figure 2 shows Δf_N for the ambient aerosol particles across various RH levels and

sampled size ranges. A negative sign in Δf_N indicates frequency reduction with increasing RH

due to water uptake. Further details of Δf_N and its derivative, $d(\Delta f_N)/(RH)$, are available in SI.

183

184 For the particle size <180 nm (Fig. 2a), Δf_N decreased, implying water uptake, with <56 nm

particles exhibiting the lowest value. In Figure 2b and 2c, particles showed increased water

uptake at lower RH, evidenced by Δf_N decrease. However, Δf_N suddenly increased at specific

187 RH, indicating a drastic water uptake leading to deliquescence-induced phase transition.

188 Figures 2a-c revealed deliquescence in particles >180 nm at different RH (known as DRH),

absent in those <180 nm. Past studies reported no deliquescence for ambient and laboratory-

190 generated atmospherically relevant particles even at high RH, attributing them to organic

191 nature (Arenas et al., 2012a; Brooks et al., 2002; Chao et al., 2020; Peng et al., 2022). We

192 hypothesize that particles <180 nm were likely dominated by freshly formed secondary

193 organic aerosols (SOA) from biogenic volatile organic compound (VOC) oxidation,

supported by the densely vegetated/forest region and season. Concurrent quasi-continuous

measurements during the same campaign indicated upto $\sim 90\%$ organic fraction in NR-PM₁

196 (non-refractory particulate matter with an aerodynamic diameter $\leq 1 \mu m$; Tab. S2) with details

197 discussed in subsequent studies.



199 Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For 200 panels a, b, and c, Δf_N represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) 201 conditions normalized to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in 202 the value of Δf_N for each size range for the sampled ambient aerosol particles indicates the water uptake at 203 different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size 204 205 ranges. In panels d, e, and f, the derivative of Δf_N with respect to RH (d(Δf_N)/d(RH)) is plotted against RH to 206 determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. 207 The RH values at which $d(\Delta f_N)/d(RH)$ becomes ≥ 1 (marked by the dotted line) represent the DRH values for the 208 individual aerosol size ranges.

209

Following Chao et al. (2020), DRH, indicated by $d(\Delta f_N)/(RH) \ge 1$ ranged from ~60 – 68% for 210 three size ranges (180 nm - 320 nm; 320 nm - 560 nm; 560 nm -1μ m) in this study (Fig. 211 2e). Above 180 nm, inorganic salts, like (NH₄)₂SO₄, may contribute to DRH appearance (Hu 212 et al., 2010) (Fig. 2b,e). These DRH values, smaller than pure (NH₄)₂SO₄ (DRH=80%), 213 suggest mixed salts and/or organic compounds in the sampled aerosol particles. The 214 deliquescence behaviour of organic-inorganic mixtures introduces complexity due to 215 solubility limitations, influenced by the organic species in the aerosol particles. Previous 216 studies noted reduced DRH of inorganic species and sometimes, unaffected by organics 217 (Smith et al., 2011, 2012, 2013). Thus, we hypothesize 180 nm to 1 µm particles at this site 218 reflect a complex mixture of inorganic salts like (NH₄)₂SO₄ and water-soluble organic 219 compounds. Supermicron particles (>1 μ m), except 1 – 1.8 μ m, showed <17% $\Delta f_{\rm N}$ indicating 220 221 lower water uptake (Fig. 2c). Beyond maximum water uptake, the three size ranges exhibited 222 different DRH values. $1.8 - 3.2 \mu m$ and $5.6 - 10 \mu m$ exhibited DRH between 56 - 68%, suggesting a mixture of water-soluble organic compounds and inorganic salts. However, 3.2 223 $-5.6 \,\mu\text{m}$ showed a very high DRH (>81%) with the organic fraction minimally affecting 224

DRH due to solubility limitations. For $1.8 - 3.2 \mu m$ and $5.6 - 10 \mu m$, reduced DRH

compared to pure inorganic salts suggests a mix of water-soluble organic and inorganic

- 227 compounds (Smith et al., 2011, 2012, 2013). The $1.0 1.8 \mu m$ range exhibited a >30% Δf_N
- reduction at ~42% RH, indicating a higher water uptake and the observed DRH appeared to
- be ~45 53%, consistent with the highest κ_m (~0.44, see discussion below) among all
- 230 measured size ranges (Chao et al., 2020). The low DRH at this site suggests presence of
- highly water-soluble organic material within the organic-inorganic (sea salt) mixture in this
- size range.
- 233
- Figure 3 shows the size-resolved mass-based hygroscopic growth factors (gf_m) and the
- corresponding mass-based hygroscopicity parameters (κ_m) of ambient aerosol particles
- calculated from QCM measurements (Text S4 and S5) over a wide RH range (2 90%). In
- Figure 3a, the growth factor increases with water uptake in humid air, showing size-
- 238 dependent variations across different RH levels (indicated by markers). Each size range had a
- 239 maximum RH limit beyond which gf_m estimation was impossible due to deliquescence, as
- observed in previous studies (Chao et al., 2020). Accordingly, the highest $gf_m=1.72$ was
- recorded at RH=90% for the 100 180 nm particles. The pronounced variations in gf_m
- 242 between different-sized particles at a given RH indicate varying chemical composition and
- 243 hygroscopicity across the sampled sizes.
- 244

Figure 3b shows the size-dependent average κ_m values, calculated from the gf_m values in 245 Figure 3a. Each particle size range exhibited minimal κ_m variation (average $\kappa_m \pm 0.007$), 246 suggesting ideal solution behaviour across the wide RH range (Z. Wang et al., 2017). The 247 248 average κ_m values (indicated by dashed lines in Figure 3b) of the size-resolved ambient particles varied significantly between 0.016 (< 0.056 nm) and 0.44 ($1.0 - 1.8 \mu$ m) over the 249 investigated RH range. This implies κ_m as a function of size, and indicates distinct chemical 250 compositions for each size fraction. For example, the small increase in κ_m from 0.016 to 251 0.078 for particles <56 nm to 56-100 nm, respectively, is potentially owing to aging 252 processes (Zhang et al., 2023). Aerosol particles <320 nm exhibited very low average κ_m 253 (0.06), which may be primarily due to the strong dominance of organic compounds in these 254 255 size ranges (Demou et al., 2003; Li et al., 2021; Shi et al., 2022). The particles in the size range of 320 nm $- 1 \mu m$ were moderately hygroscopic (average $\kappa_m = 0.19$) indicating the 256 influence of inorganic salts (Wu et al., 2016). The particles between $1.0 - 1.8 \mu m$ exhibited 257 highest κ_m (0.44), which implies the presence of highly hygroscopic material such as sea salt 258

- (Zieger et al., 2017b) mixed with organic compounds. Even larger particles exhibited reduced 259
- κ_m , possibly due to the presence of dust particles (Koehler et al., 2009), transported at this site 260
- during monsoon season (Valsan et al., 2016). The κ -Köhler parameterization fit accurately 261
- elucidates the gfm within each size range over a wide range of RH values (Petters & 262
- Kreidenweis, 2007). 263
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292 Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, 293 Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor (gfm) 294 derived using a quartz crystal microbalance (OCM) for ten different size ranges of ambient aerosol particles at 295 different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the 296 corresponding κ-Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity 297 parameters, κ_m . The error bars represent the variations in gfm averaged over the mass change corresponding to 298 different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the 299 mean κ_m values corresponding to the respective size ranges. (b) The data points are the κ_m values calculated 300 based on the gf_m (as shown in (a)) using the κ -Köhler theory for different RH conditions in the subsaturated 301 regime. The dotted lines represent the mean κ_m value for each size range of ambient aerosol particles and the 302 error bars represent one standard deviation.

303

Figure 4 shows the size dependency of κ_m in the subsaturated regime measured by QCM 304 together with $\kappa_{\rm CCN}$ (Text S2) in the supersaturated regime determined by size-resolved CCN 305 measurements and aerosol size distributions obtained by SMPS and UV-APS. The submicron 306 particles show a bimodal number size distribution with an Aitken mode peak at 63 nm and an 307 accumulation mode at 145 nm, while the mass size distributions peak at 350 nm and 3 μ m, 308

respectively. The average κ_{CCN} increased from a value of 0.28 for the Aitken mode particles 309 to 0.47 for the accumulation mode particles. In the submicron region, κ_m exhibited a nominal 310 increase with particle size (Wu et al., 2016). in parallel with the increase in κ_{CCN} , albeit at 311 much lower absolute values. This increase likely reflects the decrease of organic fraction with 312 size in the submicron range, which is supported by the appearance of deliquescence in the 313 size range above 180 nm, indicating the presence of inorganic salts. In the supermicron 314 region (for particles $\geq 1 \,\mu m$), κ_m peaked due to the potential presence of NaCl particles in the 315 size range $1.0 - 1.8 \,\mu\text{m}$, and decreased again in the higher size ranges (>1.8 μm) likely 316 317 because of the presence of dust particles. The average κ_m value (0.18) obtained from QCM measurements across all the sampled size ranges in the subsaturated regime was lower than 318 319 the average κ_{CCN} (0.39), likely due to the solubility limitation (Dusek et al., 2011; Hersey et al., 2013; Rastak et al., 2017; Riipinen et al., 2015; Wittbom et al., 2018) of organic and 320 inorganic compounds in ambient aerosol particles below the DRH. Most inorganic species 321 are completely dissolved beyond the DRH point in the subsaturated regime, allowing κ_{CCN} to 322 assume complete particle solubility (Pajunoja et al., 2015; Petters & Kreidenweis, 2007). 323



340 Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance 341 (QCM) experiments (κ_m), and from size-resolved cloud condensation nuclei (CCN) measurements (κ_{CCN}). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 -342 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass 343 344 size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by 345 assuming a density of 1.2 g cm⁻³ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements 346 347 during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived 348 from QCM experiments (κ_{m} ; orange points) and size-resolved CCN measurements (κ_{CCN} ; blue points) are shown 349 for the comparison. The error bars for κ_m and κ_{CCN} indicate the measurement uncertainty and variability, 350 respectively.

351

352 Unlike QCM measurements, the hygroscopicity parameters obtained in previous studies

using the HTDMA technique (κ_{HTDMA}) are often consistent with κ_{CCN} , as the particle

hygroscopicity is determined by particle size increase above DRH point, unaffected by 354 solubility limits (Pajunoja et al., 2015; Y. Wang et al., 2018; Wu et al., 2013). The technical 355 limitations, particularly limited diameter changes corresponding to gfm values less than 2.0, 356 prevent HTDMA from measuring aerosol hygroscopic properties below the DRH point in the 357 subsaturated regime (Laskina et al., 2015). In contrast, the QCM technique, which is not 358 subject to these limitations, can be used to understand the water uptake characteristics of 359 aerosol particles in the subsaturated regime below DRH and substantially enhance our 360 understanding about solubility limitations of organic and inorganic compounds in the lower 361 362 RH region. Such an enhancement in the knowledge is crucial for accurately quantifying the radiative forcing effects of ambient aerosol particles in the atmosphere. The strong size 363 dependence of κ_m also indicated varying chemical composition for different size ranges 364 365 measured in this study.

366 4. SUMMARY AND ATMOSPHERIC IMPLICATIONS

We report the first results based on a high-sensitivity QCM technique to investigate the mass-367 368 based growth factor and hygroscopicity parameter of size-resolved ambient aerosols over a wide range of RH from a relatively pristine high-altitude site in India. For the investigated 369 size ranges, the pronounced variations in gfm below the DRH provided an opportunity to 370 better understand the changes in ambient aerosol properties even at low RH values, which 371 may not be revealed by size-based growth factor measurements (Hu et al., 2010; Laskina et 372 al., 2015). These measurements clearly highlight the important and critical role in knowing 373 the changes in aerosol properties based on their RH history for an improved understanding of 374 water uptake, phase transition, and radiative impact of atmospheric aerosol particles (Zhao et 375 al., 2022). Based on the estimated κ_m values, the ambient aerosol particles exhibited the 376 behaviour of an ideal solution (Pajunoja et al., 2015) and strong size-dependent chemical 377 composition. We further observed a pronounced size dependency of the DRH values 378 potentially resulting from complexities of organic solubility in organic-inorganic mixtures 379 (Li et al., 2021). Such a complexity owing to the presence of various organic species may 380 381 alter the CCN behaviour of atmospheric aerosol particles due to phase transitions and changes in physical properties. The understanding of CCN activation in the supersaturated 382 regime under the assumption of complete solubility of particles is relatively well established. 383 But the understanding of the thermodynamic properties associated with particle growth in the 384 385 subsaturated regime still remains a challenge. Our findings emphasize the need for additional experiments on ambient and atmospherically relevant laboratory-generated aerosol particles 386

- using high-sensitivity techniques like QCM. We demonstrated the importance of this
- technique to better understand the rapid changes in aerosol properties resulting from exposure
- to a wide range of atmospheric RH conditions. The enhancement in our understanding of the
- 390 complex interplay between water vapour and aerosol particles will help in developing more
- accurate models to effectively describe the role of aerosols in atmospheric processes to
- reduce climate uncertainties and assess the impact of air pollution on human and ecosystem
- 393 health.

394 ASSOCIATED CONTENT

395

396 Author contributions

SSG conceived the idea. SSG and PL conceptualized the study and designed the research. CJ 397 further developed and validated the QCM measurements for the ambient aerosol studies. CJ 398 performed the field measurement campaign to collect the aerosol samples with support from 399 AS, KNK, and RKA. CJ performed all the laboratory experiments using QCM with support 400 from SS. GVJ performed the satellite data analysis to obtain the AOD values. CJ carried out 401 402 the data analysis obtained from QCM with input from PL and SSG. CJ and AS performed the scientific interpretation of the QCM data under the mentorship of RR, SSG and PL. CJ wrote 403 the first draft of manuscript under the mentorship of SSG with inputs from PL and RR, and 404 further edits from SY. MOA and STM further provided critical and valuable inputs on the 405 manuscript. 406

407

408 **Open research**

- 409 The data used in the manuscript has been deposited in an open research repository as Excel
- 410 files, accessible at https://doi.org/10.6084/m9.figshare.24512377. MODIS data were sourced
- 411 from the Level 2 and Atmosphere Archive and Distribution System (LAADS)
- 412 (https://doi.org/10.5067/MODIS/MYD04_3K.061). Figures were prepared using Igor Pro
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- 414 https://www.wavemetrics.com/software/igor-pro-9.
- 415

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Figure 1. Spatial distribution of average Aerosol Optical Depth (AOD) derived from MODIS (Moderate Resolution Imaging Spectroradiometer) Level 2 data over the Indian continental region during the monsoon season of 2021 (June - September). The AOD distribution clearly indicates a relatively lower aerosol loading over the observational site of Munnar (marked as a black triangle) compared with the other parts of India. The wind rose diagram shown in the inset is average wind speed and wind direction arriving at the sampling site during the ambient aerosol sampling period (August – September 2021). The prevailing air masses mostly originated over the Indian Ocean and arrived from southwest direction, bringing clean marine influx to the observational site, confirming the relatively low influence of anthropogenic activities.



Figure 2. Deliquescence phase transition behaviour of size-resolved ambient aerosol particles from Munnar. For panels a, b, and c, Δf_N represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the ambient aerosol particles at different relative humidity (RH) conditions normalised to that of the dry aerosol particles at RH <5%, expressed as percentage. The decrease in the value of Δf_N for each size range for the sampled ambient aerosol particles indicates the water uptake at different RH conditions in the subsaturated regime. The solid markers and lines identify different particle size ranges. In panels d, e, and f the derivative of Δf_N with respect to RH (d(Δf_N)/d(RH)) is plotted against RH to determine the deliquescence relative humidity (DRH) value corresponding to the respective aerosol size ranges. The RH values at which d(Δf_N)/d(RH) becomes ≥ 1 (marked by the dotted line) represent the DRH values for the individual aerosol size ranges.



Figure 3. Size-resolved hygroscopicity measurements of ambient aerosol particles at the high-altitude site, Munnar, during the Monsoon season (August-September 2021). (a) Mass-based hygroscopic growth factor (gf_m) derived using a quartz crystal microbalance (QCM) for ten different size ranges of ambient aerosol particles at different relative humidity (RH) conditions in the subsaturated regime (circles). The solid lines represent the corresponding κ -Köhler growth factor fits obtained using the mean value of mass-based hygroscopicity parameter, κ_m . The error bars represent the variations in gf_m averaged over the mass change corresponding to different overtone frequencies of the QCM sensor at respective RH conditions. The values in parentheses are the mean κ_m values corresponding to respective size ranges. (b) The data points are the κ_m values calculated based on the gf_m (as shown in (a)) using the κ -Köhler theory for different RH conditions in the subsaturated regime. The dotted lines represent the mean κ_m value for each size range of ambient aerosol particles and the error bars represent one standard deviation.



Figure 4. Aerosol size distributions, hygroscopicity parameters derived from quartz crystal microbalance (QCM) experiments (κ_m), and from size-resolved cloud condensation nuclei (CCN) measurements (κ_{CCN}). The number size distribution obtained using a Scanning Mobility Particle Sizer (SMPS) over the size range of 10 – 430 nm (blue curve) was measured during the sampling period and exhibited a bimodal distribution. The mass size distributions (dark grey shaded area) were derived based on the aerosol number size distribution by assuming a density of 1.2 g cm⁻³ for the submicron region. The mass size distribution for the supermicron range (light grey shaded area) was obtained using Ultraviolet Aerodynamic Particle Sizer (UV-APS) measurements during the same season (June-August) but for a different year (2014). The hygroscopicity parameters derived from QCM experiments (κ_m ; orange points) and size-resolved CCN measurements (κ_{CCN} ; blue points) are shown for the comparison. The error bars for κ_m and κ_{CCN} indicate the measurement uncertainty and variability, respectively.

Supporting Information

Complex hygroscopic behaviour of ambient aerosol particles revealed by a piezoelectric technique

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List of supporting materials:

- 1. Seven texts, Text S1 Text S7
- 2. Two tables, Table S1 Table S2
- 3. Five figures, Figure S1 Figure S5
- 4. References

Text S1. Sampling Site

The sampling was conducted at the Natural Aerosol and Bioaerosol High Altitude (NABHA) Laboratory at the College of Engineering, Munnar (10.0930° N, 77.0682° E), located in the Western Ghats of India, one of the most significant biodiversity hotspots in the world (Kale et al., 2016). The site is at about ~1600 m in elevation and 90 km from the Arabian Sea. Munnar possesses a moderately rugged topography with high mountain peaks and deep river valleys covered with diversified vegetated areas, tea plantations, barren land/rocky areas, and water bodies (Gupta et al., 2020). During the southwest monsoon season (June-September), the Western Ghats block the south-westerly winds, causing rainfall on these mountain ranges.

Text S2. Instrumentation

The ambient aerosols were collected on polytetrafluoroethylene (PTFE) membrane filters (pore size $0.2 \ \mu$ m) using a Micro-Orifice Uniform Deposit Impactor (MOUDI) with a sampling flow rate of 30 L/min. The MOUDI-II 120R (TSI) utilized 10 rotating stages to achieve a uniform deposit of particles on the filters, with nominal cut-off points at 0.056, 0.10, 0.18, 0.32, 0.56, 1.0, 1.8, 3.2, 5.6, and 10 μ m (Marple et al., 2014). Concurrently, various meteorological parameters such as temperature, humidity, pressure, wind speed, wind direction, and precipitation were recorded using an Automatic Weather Station (AWS, Clima Sensor US) at the sampling location throughout the sampling period.

The number size distribution of the ambient aerosol particles was measured using a Scanning Mobility Particle Sizer (SMPS), which comprises a 3082 Electrostatic Classifier with 3081 Long Differential Mobility Analyzer (L-DMA, TSI Model 3081) and a butanol-Condensation Particle Counter (CPC, TSI Model 3750). The measurements were carried out with an aerosol flow rate of 1 L/min, spanning a size range of 10.2 - 430 nm. The aerosol mass distribution measurements for the larger-sized particles ($0.3 - 20.0 \mu$ m) obtained using an Ultraviolet Aerodynamic Particle Sizer (UV-APS, TSI Inc., Model 3314) from a campaign conducted at

the same location during the Monsoon season of 2014 (June-August) were also incorporated in the study.

The hygroscopicity of aerosol particles was analyzed using a sensitive mass-balance Quartz Crystal Microbalance (QCM, QSense Analyzer, Biolin Scientific), which can measure the mass changes at the nanograms level. In 1959, Sauerbrey proposed that an adsorbed mass (Δm) causes a decrease in the resonant frequency (Δf) of the quartz crystal, and their linear relationship is given by,

$$\Delta f = -\frac{c}{n}\Delta m \tag{S1}$$

where C (=17.7 ng.cm⁻².Hz⁻¹ for 5 MHz AT-cut crystals) is the mass sensitivity constant specific to the quartz resonator and n (=1,3,5,...) is the overtone number (Sauerbrey, 1959). The Sauerbrey equation is valid only when the film deposited on the QCM quartz sensor is rigid and perfectly coupled to the sensor surface (Rodahl & Kasemo, 1995). QCM experiments were performed with a 5 MHz AT-cut quartz sensor coated with SiO₂, and the particles were uniformly deposited on the sensor. The ambient aerosol particles collected on PTFE filters were directly transferred to the QCM sensor by placing the filter on the sensor and then gently pressing it using a cotton piece. The particles on the sensor were examined under the microscope to ensure the uniformity of the film.

The relative humidity (RH) was incrementally increased from 2% to 93% to promote the uptake of water by aerosol particles, resulting in a corresponding increase in mass. The QCM provides the fundamental frequencies corresponding to the mass deposited on the sensor. Measurements were taken at various RH points in the sub-saturated region for a blank sensor and a sensor coated with aerosol particles. The difference in the frequencies between the coated and the blank sensor at a particular RH was subsequently converted to the mass of water absorbed or adsorbed at that point using the Sauerbrey equation.

The effective particle hygroscopicity, κ_{CCN} (Petters & Kreidenweis, 2007) under supersaturated conditions was determined using size-resolved Cloud Condensation Nuclei (CCN) measurements acquired through a CCN counter (CCNC, CCN-100, DMT), employing the method outlined in Rose et al. (2008).

Text S3. Experimental setup

The schematic experimental setup designed to measure the hygroscopicity of the aerosol particles using QCM is depicted in Figure S1. Dry nitrogen was passed through a HEPA filter (Whatman 6702-9500 HEPA-CAP filters) and divided into two channels, one of which was humidified using a water bubbler. Desired RH values were created by controlling the flow ratio of dry and humidified air from both channels using digital mass flow controllers (MFCs) (Alicat Scientific-50 SCCM). The RH was continuously monitored at various points in the flow pathway using thermohydrometers (testo 605i). The accuracy of the RH values in the ranges 10-35%, 35-65%, 65-90% and <10% or >90% are ±3.0%, ±2.0%, ±3.0% and ±5%, respectively. Each humidity condition was sustained until an equilibrium condition was reached, where no further evaporation and condensation happened. The amount of water absorbed by the particles at different values of RH in the sub-saturated region was measured continuously by the QCM. The QCM module temperature was kept at least 0.5 °C higher than the room temperature to avoid condensation of water vapour onto the particles inside the QCM module. The experimental setup employed a low flow range of 0-50 sccm, and any variations in the flow could significantly affect the experimental results. To mitigate this issue, the flow rate was checked at various control points before and after each experiment using a Gilibrator (Sensidyne).

To ensure accurate measurements, it is crucial to properly clean the sensor surface before depositing the particles. The sensors were cleaned with milli-Q water and methanol, then dried with N_2 gas. Subsequently, a UV-Ozone treatment was applied to eliminate various contaminants, particularly organic compounds, from the SiO₂-coated sensor. The treatment was done for 20 minutes at a power of 1 mW/cm² immediately before depositing the sample particles on the sensor.

Text S4. Mass-based hygroscopic growth factor (gfm) for laboratory-generated particles

The QCM technique for determining hygroscopicity was validated for a sucrose thin film. The mass-based hygroscopic growth factor (gf_m , mass ratio of the film at an elevated RH to that at <5% RH) for sucrose particles for a range of RH (2-93%) was calculated from QCM measurements, and the results were compared with the models for testing the accuracy of the method (Liu et al., 2018).

The steps followed for making a sucrose thin film on the sensor are as follows.

1. Prepared different concentrations (0.05%, 0.15, 0.25%, 0.5% etc) of sucrose solution by dissolving sucrose in milli-Q water.

2. Cleaned the sensor surface using Milli-Q water and methanol and treated with UV-Ozone for 20 minutes.

3. The desired volume of the solution was pipetted onto the sensor surface, which was placed on a clean petriplate.

4. The sensor was then placed inside a muffle furnace with a preset temperature of 80°C for 20 minutes to remove water.

5. The sensor with the film was viewed under a microscope (Dino-Lite Edge) to analyze the uniformity of the film.

6. The sensor with a thin film of sucrose was then mounted inside one of the modules of the QCM analyzer.

7. Dry air was allowed to pass over the film for 10-12 hours to further remove the moisture content in the particles.

After checking the uniformity of the film and the stability of different overtones using QSoft software (Fig. S2), the experiment was started with a preset program in FlowVision (software that helps to control the flow rates at the MFCs). The flow rates were changed accordingly in the two MFCs for creating different humidity values starting from dry conditions (<5% RH) to a maximum of 93% RH.

The hygroscopic growth factor values obtained from the QCM measurements were in good agreement with the literature (Norrish, 1966; Starzak & Peacock, 1997; Zobrist et al., 2011) (Fig. S3).

Text S5. Mass-based hygroscopicity parameter (km) for ambient aerosol particles

After transferring the aerosol particles from the filters to the sensor, the sensor was mounted inside the QCM module and the RH was incrementally increased to investigate the water uptake behavior of particles under varying humidity conditions.

We calculated the mass-based hygroscopicity parameter, κ_m , which measures the aerosol particles' ability to take up atmospheric moisture, from the growth factor values obtained from the QCM results (Petters & Kreidenweis, 2007).

$$\kappa_m = (\frac{1}{a_w} - 1)(gf_m - 1)$$
(S2)

where a_w is the water activity and gf_m is the mass-based growth factor of the particle at different RH values. The a_w was approximated as the fractional RH under equilibrium conditions. κ_m could be converted to a volume-based hygroscopicity parameter κ_v , using the following relationship.

$$\kappa_{\nu} = \kappa_m \times \frac{\rho_d}{\rho_w} \tag{S4}$$

where ρ_d is the effective density of the dry particle and ρ_w is the density of water (Mikhailov et al., 2013).

The experiment was repeated for the aerosol particles from different stages of MOUDI to carry out a size-resolved study. In this study, the ambient particles were solid and rigid until they deliquesce at higher RH conditions. The QCM can measure the mass of the particles under different RH conditions till they reach the phase transition point (deliquescence relative humidity, DRH).

Text S6. Estimation of Deliquescence Relative Humidity (DRH) and Aerosol Liquid Water Content (ALWC)

The phase transition behavior of ambient aerosols was investigated by analyzing the QCM frequency response curves. During the humidification cycle, the DRH point is identified as the RH at which the oscillation frequencies of QCM response start increasing upon the addition of water (Arenas et al., 2012; Chao et al., 2020). A percentage value, Δf_N , is obtained by normalizing the oscillation frequency change induced by the aerosol particles at a given RH (Δf_R) to the oscillation frequency change resulting from a dry sample (Δf_S at RH <5%) (Chao et al., 2020).

$$\Delta f_N = \left(\frac{\Delta f_R}{\Delta f_S}\right) \times (-100)\% \tag{S4}$$

The negative sign indicates that the oscillation frequency of the sensor decreases when RH increases due to water uptake by the aerosol particles. Δf_N and its derivative with respect to the measured RH, $d(\Delta f_N)/d(RH)$ were plotted for different size ranges of ambient aerosol samples (Fig. 3). The deliquescence of the particles is indicated by an increase in Δf_N or a positive value of $d(\Delta f_N)/d(RH)$. The RH at which $d(\Delta f_N)/d(RH) \ge 1$ is considered as the DRH point for the respective sample (Chao et al., 2020). Following this method, the DRH for (NH₄)₂SO₄ particles was estimated between 76-80% from the QCM measurements (Fig. S4), which agrees with the literature values (Arenas et al., 2012; Martin, 2000; Peng et al., 2022).

Similarly, the DRH values for size-resolved ambient aerosol particles from Munnar were estimated.

Aerosol Liquid Water Content (ALWC) at different RH conditions were also estimated from the size-resolved hygroscopic growth factor values combined with the particle number size distribution (PNSD) measurements, assuming volume conservation during the aerosol hygroscopic process (Bian et al., 2014).

$$ALWC = \left[\frac{\pi}{6}\sum_{i} N_i D_{d,i}^3 (gf(D_d, RH)^3 - 1)\right] \rho_W$$
(S5)

where N_i represents the number concentration of dry particles of the *i* th bin obtained from SMPS and UV-APS measurements, $D_{d,i}$ is the particle diameter, and ρ_W is the density of water.

Text S7. Correction for the RH values inside the QCM module

The flow rate of humidified air in the setup can be impacted by temperature instabilities, potentially affecting the results. To prevent water vapor condensation, it is crucial to maintain the QCM module temperature at a level equal to or higher than the room temperature. To account for any temperature changes inside the QCM module, the RH was corrected using the Clausius-Clapeyron equation.

$$ln\frac{P_1}{P_2} = \frac{\Delta H_{vap}}{R} \left(\frac{1}{T_2} - \frac{1}{T_1}\right)$$
(S6)

where P_1 and P_2 are the vapor pressures at two temperatures, T_1 and T_2 , ΔH_{vap} is the enthalpy (heat) of vaporization, and *R* is the gas constant (8.3145 J mol⁻¹ K⁻¹). The actual RH value was determined by computing the ratio of actual vapor pressure to saturation vapor pressure, expressed as a percentage.

Table S1. Mass-based hygroscopic growth factor (gf_m) and the corresponding hygroscopicity parameter (κ_m) at different RH values in the subsaturated region for different size ranges of ambient aerosol particles (<10 μ m) sampled from Munnar using MOUDI. The measurements were obtained from the QCM experiments.

Particle size range (µm)											
<0.056				0.056-0.1			0.1-0.18				
RH (%)	gf_m	κ _m	$\kappa_{\rm v}$	RH (%)	gf_{m}	κ _m	$\kappa_{\rm v}$	RH (%)	gf_{m}	κ _m	$\kappa_{\rm v}$
2.87	1.000			5.02	1.000			3.45	1.000		
11.47	1.004	0.033	0.040	16.90	1.011	0.054	0.064	16.86	1.018	0.088	0.105
27.93	1.009	0.008	0.010	31.96	1.026	0.055	0.065	31.77	1.025	0.054	0.064
43.74	1.015	0.020	0.024	44.03	1.054	0.069	0.082	44.37	1.050	0.062	0.075
58.31	1.020	0.014	0.017	54.65	1.089	0.074	0.088	54.88	1.068	0.056	0.067
66.73	1.027	0.013	0.016	64.93	1.145	0.078	0.094	66.04	1.098	0.051	0.061
72.97	1.034	0.013	0.015	73.29	1.282	0.103	0.124	75.49	1.189	0.062	0.074
80.74	1.051	0.012	0.015	79.76	1.426	0.108	0.130	90.15	1.722	0.079	0.095
				86.56	1.549	0.085	0.102	86.85	1.567	0.086	0.103

Particle size range (µm)											
0.18-0.32				0.32-0.56			0.56-1.0				
RH (%)	gf_m	κ _m	$\kappa_{\rm v}$	RH (%)	gf_m	κ _m	$\kappa_{\rm v}$	RH (%)	gf_m	κ _m	$\kappa_{\rm v}$
2.95	1.000			3.47	1.000			3.09	1.000		
12.42	1.008	0.059	0.071	12.26	1.027	0.194	0.232	12.34	1.030	0.210	0.251
21.84	1.025	0.090	0.109	22.58	1.055	0.188	0.226	21.50	1.056	0.203	0.244
31.36	1.038	0.083	0.099	30.21	1.079	0.182	0.218	30.57	1.079	0.179	0.214
47.70	1.055	0.060	0.072	47.33	1.158	0.175	0.211	46.91	1.149	0.169	0.203
55.62	1.071	0.057	0.068	55.83	1.232	0.183	0.220	55.25	1.220	0.178	0.214
64.23	1.145	0.081	0.097	64.13	1.365	0.204	0.245	63.93	1.369	0.208	0.250

Particle size range (µm)											
1.0-1.8				1.8-3.2				3.2-5.6			
RH (%)	gf_m	κ _m	$\kappa_{\rm v}$	RH (%)	gfm	κ _m	$\kappa_{\rm v}$	RH (%)	gf_m	κ _m	$\kappa_{\rm v}$
3.74	1.000			2.80	1.000			3.33	1.000		
12.14	1.061	0.438	0.526	12.39	1.027	0.191	0.229	11.63	1.013	0.097	0.116
19.58	1.110	0.451	0.541	21.64	1.053	0.191	0.229	19.70	1.029	0.119	0.143
27.80	1.167	0.435	0.522	31.09	1.100	0.223	0.267	28.52	1.053	0.134	0.161
35.00	1.244	0.452	0.543	39.25	1.145	0.225	0.270	35.55	1.056	0.101	0.121
42.85	1.315	0.421	0.505	47.91	1.180	0.195	0.234	43.24	1.075	0.099	0.119
47.37	1.411	0.457	0.548					50.94	1.123	0.119	0.142
								59.05	1.180	0.125	0.150

Particle size range (µm)								
	5.6-10.0							
RH (%)	gf _m	κ _m	κ _v					
3.14	1.000							
12.25	1.015	0.107	0.128					
21.61	1.030	0.110	0.132					
31.22	1.046	0.101	0.121					
39.43	1.067	0.103	0.123					
48.00	1.131	0.142	0.171					
56.41	1.177	0.137	0.164					
64.77	1.198	0.108	0.129					

Table S2. The average mass concentration (μ g m⁻³) and the mass fraction of the chemical species in NR-PM₁ measured by aerosol chemical speciation monitor (ACSM) during the period from 06.06.2021 to 27.07.2021.

NR-PM ₁ Species	Mass concentration (µg m ⁻³)	Mass fraction
Organics	1.15	0.50
Sulphate	0.79	0.35
Nitrate	0.08	0.04
Ammonium	0.22	0.09
Chloride	0.04	0.02



Figure S1. Schematic diagram of the experimental setup for providing a wide range of humidity conditions (2-93%) to study the aerosol-water vapor interactions of the samples coated on the quartz crystal microbalance (QCM) sensor mounted inside the module of the chamber platform. Different humidity conditions were created over the aerosol particles by controlling the flowrates of dry and humidified air using the two mass flow controllers (MFCs) connected in parallel. The RH of incoming and outgoing air is measured using two thermohygrometers connected before and after the QCM chamber platform.



Figure S2. The screenshot of the QSoft software showing the oscillation frequency changes of the quartz crystal microbalance (QCM) sensor with time during the humidifying and drying phase of a QCM experiment for the size ranges 56-100 nm and 560 nm-1 μ m of ambient aerosol particles from Munnar. (a) shows a decrease in the oscillation frequency change for the particles in the size range 56-100 nm, corresponding to an increase in mass due to water uptake during the humidifying (increasing RH) phase and a further increase in the frequency change when the water content is removed during the drying (decreasing RH) phase. (b) shows an initial decrease in the frequency change due to water uptake by the aerosol particles in the size range 560 nm-1 μ m in the lower RH region (<61.2%) during the humidification cycle. The increase in frequency changes and the separation of overtones at RH=61.2% in the humidifying phase indicate that the particles undergo phase transition (deliquescence) at high RH region and become aqueous droplets leading to less viscous and non-rigid particles on the sensor. The initial frequency change value corresponding to the dry particle mass is obtained after the humidification and drying phase, indicating no particle loss. Different curves correspond to different overtone frequencies of the sensor.



Figure S3. Mass-based hygroscopic growth factor gf_m of amorphous sucrose particles at different RH values in the subsaturated region obtained from the quartz crystal microbalance (QCM) measurements. The experimental values strictly follow the gf_m values retrieved using different models (Norrish, Starzak and Peacock, and Zobrist) from the literature.



Figure S4. The deliquescence phase transition behavior of laboratory-generated (NH₄)₂SO₄ particles. Δf_N represents the change in the oscillation frequency of the quartz crystal microbalance (QCM) sensor resulting due to water uptake by the particles at different RH conditions normalized to that of the dry particles at RH <5%, expressed as percentage. The decrease in the value of Δf_N indicates the water uptake at different RH conditions in the subsaturated region. The derivative of Δf_N with respect to RH, $d(\Delta f_N)/d(RH)$ is plotted against different RH conditions and the respective RH values for $d(\Delta f_N)/d(RH) \ge 1$, marked by the black dotted line, represent the deliquescence relative humidity (DRH) range (76-80%) for the (NH₄)₂SO₄ particles.



Figure S5. Aerosol Liquid Water Content (ALWC) in size-resolved ambient aerosol particles from Munnar across varied relative humidity (RH) conditions in the subsaturated regime. ALWC under different RH conditions is estimated by integrating size-resolved hygroscopic growth factor values with particle number size distribution (PNSD) measurements, assuming volume conservation during aerosol water uptake processes. The marker sizes correspond linearly to particle size, while marker colors signify the mean value of the mass-based hygroscopicity parameter (κ_m) for each size-range, indicated by the color palette. The total ALWC averaged over all particles below 10 μ m in size is also shown in inset.

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